

# PATENT COOPERATION TREATY

## PCT

### NOTIFICATION OF ELECTION

(PCT Rule 61.2)

From the INTERNATIONAL BUREAU

To:

Assistant Commissioner for Patents  
United States Patent and Trademark  
Office  
Box PCT  
Washington, D.C. 20231  
ETATS-UNIS D'AMERIQUE

in its capacity as elected Office

<b>Date of mailing (day/month/year)</b> 10 August 2000 (10.08.00)	
<b>International application No.</b> PCT/US99/24698	<b>Applicant's or agent's file reference</b> BB1255 PCT
<b>International filing date (day/month/year)</b> 21 October 1999 (21.10.99)	<b>Priority date (day/month/year)</b> 23 October 1998 (23.10.98)
<b>Applicant</b> ABELL, Lynn, Marie et al	

1. The designated Office is hereby notified of its election made:

☒ in the demand filed with the International Preliminary Examining Authority on:

28 April 2000 (28.04.00)

☐ in a notice effecting later election filed with the International Bureau on:

2. The election ☒ was  
☐ was not

made before the expiration of 19 months from the priority date or, where Rule 32 applies, within the time limit under Rule 32.2(b).

<b>The International Bureau of WIPO</b> 34, chemin des Colombettes 1211 Geneva 20, Switzerland	<b>Authorized officer</b> <p style="text-align: center;">Maria Kirchner</p>
Facsimile No.: (41-22) 740.14.35	Telephone No.: (41-22) 338.83.38



## PATENT COOPERATION TREATY

RECEIVED

JAN 29 2001

From the INTERNATIONAL SEARCHING AUTHORITY

PCT

PATENT RECORDS  
CENTER

To:  
E.I. DU PONT DE NEMOURS AND COMPANY  
Legal/Patent Records Center  
Attn. ~~FEULNER, G.~~  
1007 Market Street  
Wilmington, Delaware 19898  
UNITED STATES OF AMERICA

TMR

## INVITATION TO PAY ADDITIONAL FEES

(PCT Article 17(3)(a) and Rule 40.1)

JAN 31 2001

CC: AG BIOTECH

Applicant's or agent's file reference

BB1255 PCT

Date of mailing  
(day/month/year)

16/01/2001

## PAYMENT DUE

within 45 ~~XXXX~~ days  
from the above date of mailing

International application No.

PCT/US 99/24698

International filing date  
(day/month/year)

21/10/1999

Applicant

E.I. DU PONT DE NEMOURS AND COMPANY et al.

## 1. This International Searching Authority

- (i) considers that there are 4 (number of) inventions claimed in the international application covered by the claims indicated ~~XXXX~~ on the extra sheet:

and it considers that the international application does not comply with the requirements of unity of invention (Rules 13.1, 13.2 and 13.3) for the reasons indicated ~~XXXX~~ on the extra sheet:

- (ii) ☒ has carried out a partial international search (see Annex) ☐ will establish the international search report on those parts of the international application which relate to the invention first mentioned in claims Nos.:  
**1-15, all partially**
- (iii) will establish the international search report on the other parts of the international application only if, and to the extent to which, additional fees are paid

## 2. The applicant is hereby invited, within the time limit indicated above, to pay the amount indicated below:

EUR 945,00 x 3 = EUR 2.835,00  
Fee per additional invention      number of additional inventions      total amount of additional fees

Or, \_\_\_\_\_ x \_\_\_\_\_ = \_\_\_\_\_

The applicant is informed that, according to Rule 40.2(c), the payment of any additional fee may be made under protest, i.e., a reasoned statement to the effect that the international application complies with the requirement of unity of invention or that the amount of the required additional fee is excessive.

3. ☐ Claim(s) Nos. \_\_\_\_\_ have been found to be unsearchable under Article 17(2)(b) because of defects under Article 17(2)(a) and therefore have not been included with any invention.

Name and mailing address of the International Searching Authority



European Patent Office, P.B. 5818 Patentlaan 2  
NL-2280 HV Rijswijk  
Tel. (+31-70) 340-2040, Tx: 31 651 epo nl,  
Fax: (+31-70) 340-3016

Authorized officer

Renate Jordan

REY NOTED



1. The present communication is an Annex to the invitation to pay additional fees (Form PCT/ISA/206). It shows the results of the international search established on the parts of the international application which relate to the invention first mentioned in claims Nos.:  
see 'Invitation to pay additional fees'
2. This communication is not the international search report which will be established according to Article 18 and Rule 43.
3. If the applicant does not pay any additional search fees, the information appearing in this communication will be considered as the result of the international search and will be included as such in the international search report.
4. If the applicant pays additional fees, the international search report will contain both the information appearing in this communication and the results of the international search on other parts of the international application for which such fees will have been paid.

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	FUJIMORI K ET AL: "Molecular cloning and characterization of the gene encoding N'-((5'-phosphoribosyl)-formimino)-5-amino imidazole-4-carboxamide ribonucleotide (BBM 11) isomerase from Arabidopsis thaliana." MOLECULAR & GENERAL GENETICS, vol. 259, no. 2, August 1998 (1998-08), pages 216-223, XP000960744 ISSN: 0026-8925 the whole document ---	1-8, 10-15
Y	FANI R ET AL: "Paralogous histidine biosynthetic genes: evolutionary analysis of the Saccharomyces cerevisiae HIS6 and HIS7 genes" GENE: AN INTERNATIONAL JOURNAL ON GENES AND GENOMES, GB, ELSEVIER SCIENCE PUBLISHERS, BARKING, vol. 197, no. 1-2, 15 September 1997 (1997-09-15), pages 9-17, XP004126397 ISSN: 0378-1119 the whole document. --- -/--	1-8, 10-15

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

\* Special categories of cited documents:

- \*A\* document defining the general state of the art which is not considered to be of particular relevance
- \*E\* earlier document but published on or after the international filing date
- \*L\* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- \*O\* document referring to an oral disclosure, use, exhibition or other means
- \*P\* document published prior to the international filing date but later than the priority date claimed

- \*T\* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- \*X\* document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- \*Y\* document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- \*B\* document member of the same patent family



C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	JEANMOUGIN F ET AL: "Multiple sequence alignment with Clustal X" TIBS TRENDS IN BIOCHEMICAL SCIENCES, EN, ELSEVIER PUBLICATION, CAMBRIDGE, vol. 23, no. 10, 1 October 1998 (1998-10-01), pages 403-405, XP004146847 ISSN: 0968-0004 the whole document	1-8, 10-15
A	--- BORK PEER ET AL: "Divergent evolution of a beta/alpha-barrel subclass: Detection of numerous phosphate-binding sites by motif search." PROTEIN SCIENCE, vol. 4, no. 2, 1995, pages 268-274, XP000965754 ISSN: 0961-8368 the whole document	1,2,10
A	--- WO 92 19629 A (JAPAT LTD ;CIBA GEIGY JAPAN LTD (JP)) 12 November 1992 (1992-11-12) abstract paragraphs '0001!', '0002!', sentence 1 page 7-8 page 24, paragraph 2	1,11-13
A	--- CRANE D I ET AL: "The Pichia pastoris HIS4 gene: nucleotide sequence, creation of a non-reverting his4 deletion mutant, and development of HIS4-based replicating and integrating plasmid" CURRENT GENETICS, NEW YORK, NY, US, vol. 26, 1994, pages 443-450, XP002116157 ISSN: 0172-8083 the whole document	1-8,14, 15
A	--- FANI RENATO ET AL: "Molecular evolution of the histidine biosynthetic pathway." JOURNAL OF MOLECULAR EVOLUTION, vol. 41, no. 6, 1995, pages 760-774, XP002153543 ISSN: 0022-2844 abstract; figures 1,2; table 1	1
P,X	--- DATABASE EMBL NUCLEOTIDE AND PROTEIN SEQUENCES, Hinxton, GB 29-07-99 AC = AI899888. Glycine max cDNA clone GENOME SYSTEMS CLONE ID:Gm-c1017-1135 5' similar to TR:082782 (5'-ProFAR); mRNA sequence; EST. XP002153544 abstract	1,14,15
	--- -/--	



1



ex to Form PCT/ISA/206  
COMMUNICATION RELATING TO THE RESULTS  
OF THE PARTIAL INTERNATIONAL SEARCH

International Application No  
PCT/US 99/24698

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
P, X	<p>DATABASE EMBL NUCLEOTIDE AND PROTEIN SEQUENCES, Hinxton, GB 28-07-99 AC = AI901450. 618004D06.x1 618 - Inbred Tassel cDNA Library Zea mays cDNA, mRNA sequence. EST. XP002153545 abstract</p> <p>-----</p>	1, 14, 15



This International Searching Authority found multiple (groups of) inventions in this international application, as follows:

1. Claims: 1-15, all partially

A composition consisting of an isolated polynucleotide comprising a nucleotide sequence selected from the group of SEQ.ID.N. 1-13 and their corresponding encoded polypeptides: SEQ.ID.N. 2-14; a chimeric gene; a host cell; a viral vector; a method of detecting and isolating such polynucleotides and polypeptides.

2. Claims: 1-15, all partially

A composition consisting of an isolated polynucleotide comprising a nucleotide sequence according to SEQ.ID.N. 15 and its corresponding encoded polypeptide: SEQ.ID.N. 16; a chimeric gene; a host cell; a viral vector; a method of detecting and isolating such polynucleotides and polypeptides.

3. Claims: 1-15, all partially

A composition consisting of an isolated polynucleotide comprising a nucleotide sequence selected from the group of SEQ.ID.N. 17 and 19 and their corresponding encoded polypeptides: SEQ.ID.N. 18 and 20; a chimeric gene; a host cell; a viral vector; a method of detecting and isolating such polynucleotides and polypeptides.

4. Claims: 1-15, all partially

A composition consisting of an isolated polynucleotide comprising a nucleotide sequence according to SEQ.ID.N. 21 and its corresponding encoded polypeptide: SEQ.ID.N. 22; a chimeric gene; a host cell; a viral vector; a method of detecting and isolating such polynucleotides and polypeptides.

Motivation of lack of unity

The application lacks unity of invention as required by Art.3(4)(iii) and 17(3)(a) PCT for the following reasons:

The common inventive concept underlying the application can be defined as the provision of isolated polynucleotides encoding plant phosphoribosyl-formimino)-5-aminoimidazole-4-carboxamide ribonucleotide (or ribotide) isomerase (5'-ProFAR isomerase).

The document of Fujimori et al., 1998 (Mol. Gen. Genet, 259: 216-223) describes the molecular cloning and characterization of Arabidopsis thaliana 5'-ProFAR isomerase.

In view of this prior art, the above mentioned common concept is not



novel anymore and thus, the problem underlying the application can be redefined as the provision of further isolated polynucleotides encoding plant 5'-ProFAR isomerases.

Since the single general concept is not novel the requirement of Rule 13.1 is not fulfilled, and hence, there is lack of unity. Neither the description, nor the claims revealed any further features that could be considered special in the sense of Rule 13.2 PCT. the polynucleotides disclosed in present application represent 5'-ProFAR isomerases isolated from different plants.

It is thus concluded that no technical relationship is established involving novel special technical features in the sense of rule 13.2 PCT. In consequence, the groups of inventions, i.e. inventions 1-4 are not so linked as to form a single general inventive concept as required by Article 17(3)(a) and Rule 13.1 PCT.

It is to be noted that for regrouping the different inventions, the ISA was taking into account the balance between necessary search effort and the levying of additional fees.



**Patent Family Annex**

Information on patent family members

International Application No

PCT/US 99/24698

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
WO 9219629	A	12-11-1992	AU 1554492 A	21-12-1992
			EP 0636135 A	01-02-1995
			JP 6507156 T	11-08-1994
<hr/>				

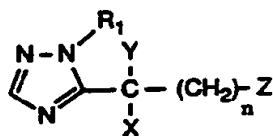






## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

<b>(51) International Patent Classification <sup>5</sup> :</b>  <b>C07F 9/6518, A01N 57/00</b>	<b>A1</b>	<b>(11) International Publication Number:</b> <b>WO 92/19629</b>  <b>(43) International Publication Date:</b> 12 November 1992 (12.11.92)
<b>(21) International Application Number:</b> PCT/JP92/00485 <b>(22) International Filing Date:</b> 17 April 1992 (17.04.92) <b>(30) Priority data:</b> 3/125510 27 April 1991 (27.04.91) JP <b>(71) Applicant (for JP only):</b> CIBA-GEIGY (JAPAN) LIMITED [JP/JP]; 10-66, Miyuki-cho, Takarazuka-shi, Hyogo 652 (JP). <b>(71) Applicant (for all designated States except JP US):</b> JAPAT LTD. [CH/CH]; Klybeckstrasse 141, CH-4057 Basle (CH). <b>(72) Inventors; and</b> <b>(75) Inventors/Applicants (for US only) :</b> MORI, Ichiro [JP/JP]; 18-31-803, Higashiashiya-cho, Ashiya-shi, Hyogo 652 (JP). IWASAKI, Genji [JP/JP]; 1-6-25-103, Mukoyama-cho, Takarazuka-shi, Hyogo 652 (JP). SCHEIDEGGER, Alfred [CH/CH]; Via Tosello, CH-6946 Ponte Capriasca (CH). KOIZUMI, Shinichi [JP/JP]; 3-5-813, Komyo-cho, Takarazuka-shi, Hyogo 652 (JP). HAYAKAWA, Kenji [JP/JP]; 1-11-4-1213, Sakasedai, Takarazuka-shi, Hyogo 652 (JP). MANO, Junichi [JP/JP]; 19-6, Hinode-cho, Suita-shi, Osaka 530 (JP).		<b>(74) Agents:</b> HANABUSA, Tsuneo et al.; Hanabusa Patent Office, Ochanomizu Square Building, 6, Kandasurugadai 1-chome, Chiyoda-ku, Tokyo 101 (JP).  <b>(81) Designated States:</b> AT (European patent), AU, BE (European patent), BR, CA, CH (European patent), DE (European patent), DK (European patent), ES (European patent), FR (European patent), GB (European patent), GR (European patent), IT (European patent), JP, KR, LU (European patent), MC (European patent), NL (European patent), RU, SE (European patent), US.  <b>Published</b> <i>With international search report.</i>

**(54) Title:** TRIAZOLE COMPOUNDS

(1)

**(57) Abstract**

This invention relates to a novel triazole compound represented by formula (1) wherein R<sub>1</sub> represents a hydrogen atom or a group A which is a protective group or is C<sub>1</sub>-C<sub>4</sub>-alkyl; X and Y jointly represent a carbonyl group together with the carbon atom to which they are bonded, or X and Y each independently represent a hydrogen atom or a -OR<sub>2</sub> group, wherein R<sub>2</sub> represents a hydrogen atom, a lower alkyl group, an acetyl group, a benzyl group or a silyl group represented by -SiR'<sub>3</sub> (wherein R' represents an alkyl group); and Z represents a -CH<sub>2</sub>PO(OR<sub>3</sub>)<sub>2</sub> or -CH<sub>2</sub>OPO(OR<sub>3</sub>)<sub>2</sub> group wherein R<sub>3</sub> represents a hydrogen atom, a lower alkyl group, a benzyl group, a phenyl group, a cyanoethyl group, an allyl group, a CH<sub>2</sub>OH group, a COOR<sub>5</sub> group, or an alkali metal, alkaline earth metal, ammonium, organic ammonium, trialkylsulphonium, trialkylsulfoxonium, phosphonium or amidinium cation; n is 0 or 1; and R<sub>5</sub> is a C<sub>1</sub>-C<sub>6</sub>-alkyl group, a process for preparing the same, a composition containing the same, as well as, application of the compound in the technical fields of herbicide or microbicides.

**FOR THE PURPOSES OF INFORMATION ONLY**

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AT	Austria	FI	Finland	MI	Mali
AU	Australia	FR	France	MN	Mongolia
BB	Barbados	GA	Gabon	MR	Mauritania
BE	Belgium	GB	United Kingdom	MW	Malawi
BF	Burkina Faso	GN	Guinea	NL	Netherlands
BG	Bulgaria	GR	Greece	NO	Norway
BJ	Benin	HU	Hungary	PL	Poland
BR	Brazil	IE	Ireland	RO	Romania
CA	Canada	IT	Italy	RU	Russian Federation
CF	Central African Republic	JP	Japan	SD	Sudan
CG	Congo	KP	Democratic People's Republic of Korea	SE	Sweden
CH	Switzerland	KR	Republic of Korea	SN	Senegal
CI	Côte d'Ivoire	LI	Liechtenstein	SU	Soviet Union
CM	Cameroon	LK	Sri Lanka	TD	Chad
CS	Czechoslovakia	LU	Luxembourg	TG	Togo
DE	Germany	MC	Monaco	US	United States of America
DK	Denmark	MG	Madagascar		
ES	Spain				

## DESCRIPTION

Triazole compounds

This invention relates to novel triazole compounds, processes for preparing the same, compositions containing the same, as well as, applications of the compounds in the technical fields of herbicides and microbicides.

Compounds of various chemical structures having herbicidal or microbicidal action have so far been developed to provide various and particularly selective herbicides and microbicides. Under such circumstances, more emphasis is being put on the protection of environment, and development of such herbicides and microbicides as do not cause environmental disruption neither secondarily nor later is desired.

Triazole compounds which have a herbicidal action are generally known. For example, European Patent Application No. 0 078 613 describes herbicidally active triazole compounds.

First of all, this invention relates to novel compounds represented by formula (1):



wherein  $\text{R}_1$  represents a hydrogen atom or a group A which is a protective group or is  $\text{C}_1$ - $\text{C}_4$ -alkyl; X and Y jointly represent a carbonyl group together with the carbon atom to which they are bonded, or X and Y each independently represent a hydrogen atom or a - $\text{OR}_2$  group, wherein  $\text{R}_2$  represents a hydrogen atom, a lower alkyl group, an acetyl group, a benzyl group or a silyl group represented by  $-\text{SiR}'_3$  (wherein  $\text{R}'$  represents an alkyl group); and Z represents a  $-\text{CH}_2\text{PO}(\text{OR}_3)_2$  or  $-\text{CH}_2\text{OPO}(\text{OR}_3)_2$  group (wherein  $\text{R}_3$  represents a hydrogen atom, a lower alkyl group, a benzyl group, a phenyl group, a cyanoethyl group, an allyl group, a  $\text{CH}_2\text{OH}$  group, a  $\text{COOR}_5$  group, or an alkali metal, alkaline earth metal, ammonium, organic ammonium, trialkylsulphonium,

- 2 -

trialkylsulfoxonium, phosphonium or amidinium cation;

n is 0 or 1; and

R<sub>5</sub> is a C<sub>1</sub>-C<sub>6</sub>-alkyl group.

This invention includes all of the isomers represented by the following formula of resonance structure:



in which R<sub>1</sub>, n, X, Y and Z have the meanings as defined in formula (1).

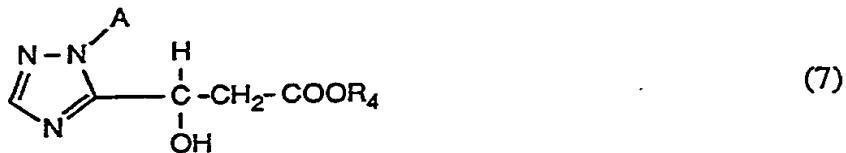
The present compound includes two kinds of compounds; one is phosphate compounds and the other phosphonate compounds.

Alkyl is, for example, methyl, ethyl, isopropyl, n-propyl, n-butyl, isobutyl, sec-butyl, tert-butyl and the various isomeric pentyl or hexyl radicals. The term "lower alkyl group" is preferably methyl, ethyl, isopropyl, n-propyl, n-butyl, isobutyl, sec-butyl or tert-butyl.

The meaning -SiR'<sub>3</sub> (wherein R' represents an alkyl group) also includes, for example, -Si(CH<sub>3</sub>)<sub>3</sub>, Si(C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>CH<sub>3</sub> or Si(CH<sub>3</sub>)<sub>2</sub>-CH(CH<sub>3</sub>)<sub>3</sub>.

The compounds represented by formula (1), wherein n is 1 and Z represents -CH<sub>2</sub>OPO(OR<sub>3</sub>)<sub>2</sub>, can be prepared by:

forming a ((1,2,4)-triazol-5-yl)propionic acid ester represented by formula (7):



wherein R<sub>4</sub> represents an alkyl group; and A has the same meaning as defined below,

- 3 -

from a ((1,2,4)-triazol-5-yl)aldehyde represented by formula (6):

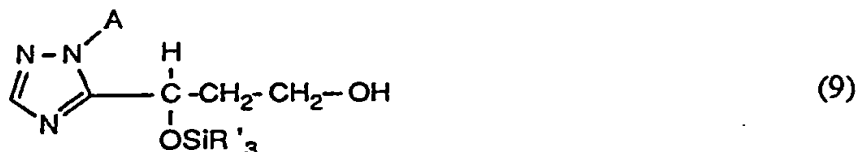


wherein A represents a protective group;

reacting the ((1,2,4)-triazol-5-yl)propionic acid ester thus formed with a suitable alkylsilyl halide to form a 3-alkylsilyloxy-3-((1,2,4)-triazol-5-yl)propionic acid ester represented by formula (8):

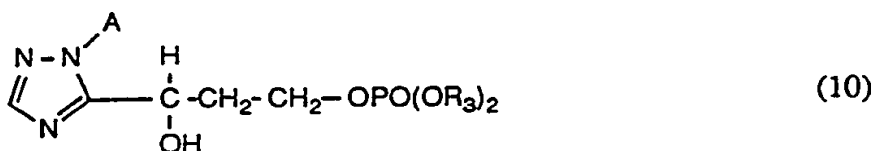


wherein R' represents an alkyl group; and A and R<sub>4</sub> have the same meanings as defined above, reducing the 3-alkylsilyloxy-3-((1,2,4)-triazol-5-yl)-propionic acid ester to form a 3-alkylsilyloxy-3-((1,2,4)-triazol-5-yl)-1-propanol represented by the formula (9):



wherein R' and A have the same meanings as defined above;

reacting the 3-alkylsilyloxy-3-((1,2,4)-triazol-5-yl)-1-propanol with a suitable phosphine compound followed by desilylation to form a 3-hydroxy-3-((1,2,4)-triazol-5-yl)propyl-phosphate represented by formula (10):

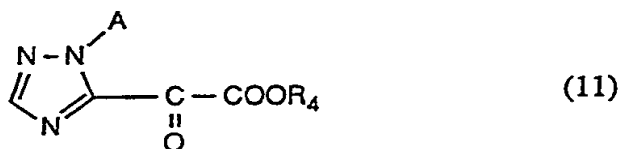


- 4 -

wherein  $R_3$  and A have the same meaning as defined above; and converting, as necessary, the 3-hydroxy-3-((1,2,4)-triazol-5-yl)propyl-phosphate to 3-hydroxy-3-((1,2,4)-triazol-5-yl)propyl-phosphate (mono- or tri-ester form).

The compounds of the formula (1), wherein n is 0 and Z represents  $-\text{CH}_2\text{OPO}(\text{OR}_3)_2$  can be prepared by:

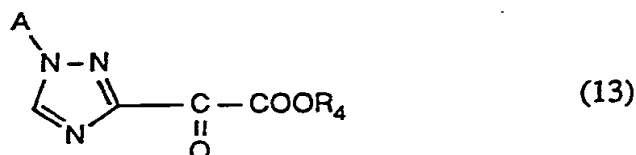
forming a 2-((1,2,4)-triazol-5-yl)-2-oxoacetic acid ester represented by formula (11):



wherein  $R_4$  represents an alkyl group; and A has the same meaning as defined below, from a (1,2,4)-triazole represented by formula (12):

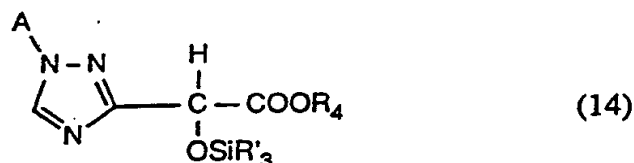


wherein A represents a group which is a protective group or is  $\text{C}_1\text{--C}_4$ -alkyl; isomerizing the 2-((1,2,4)-triazol-5-yl)-2-oxoacetic acid ester thus formed to form 2-((1,2,4)-triazol-3-yl)-2-oxoacetic acid ester represented by formula (13):

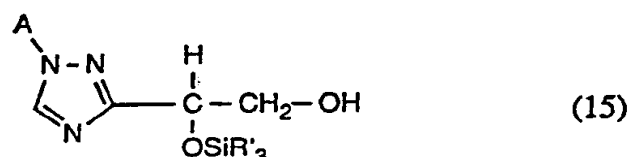


wherein A and  $R_4$  have the same meanings as defined above; reducing the 2-((1,2,4)-triazol-3-yl)-2-oxoacetic acid ester followed by silylation to form 2-((1,2,4)-triazol-3-yl)-2-alkylsilyloxy-acetic acid ester represented by formula (14):

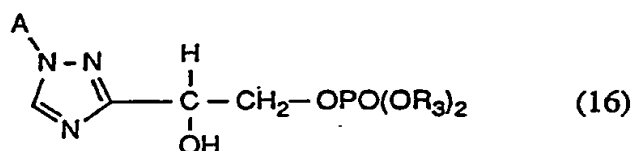
- 5 -



wherein R' represents an alkyl group; and A and R<sub>4</sub> have the same meanings as defined above, reducing the 2-alkylsilyloxy-2-((1,2,4)-triazol-3-yl)acetic acid ester to form a 2-alkylsilyloxy-2-((1,2,4)-triazol-3-yl)-1-ethanol represented by formula (15):



wherein R' and A have the same meanings as defined above; reacting the 2-alkylsilyloxy-2-((1,2,4)-triazol-3-yl)-1-ethanol with a suitable phosphine compound followed by desilylation to form a 2-hydroxy-2-((1,2,4)-triazol-3-yl)ethyl-phosphate represented by formula (16):



wherein A and R<sub>3</sub> have the same meaning as defined above; and converting, as necessary, the 2-hydroxy-2-((1,2,4)-triazol-3-yl)ethyl-phosphate to 2-hydroxy-2-((1,2,4)-triazol-3-yl)-ethyl-phosphate (mono or triester form); whereas the compounds of formula (1), wherein n is 1 and Z represents -CH<sub>2</sub>PO(OR<sub>3</sub>)<sub>2</sub> can be prepared by

reacting a (1,2,4)-triazole represented by formula (12):



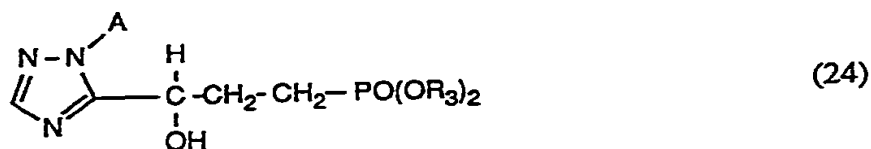
wherein A represents a group which is a protective group or is C<sub>1</sub>-C<sub>4</sub>-alkyl; with an

- 6 -

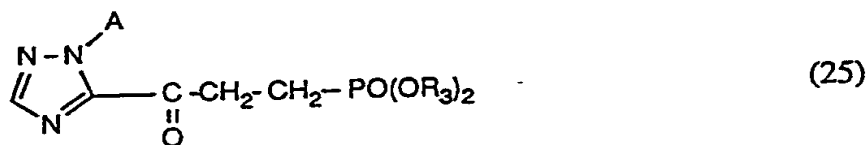
aldehyde compound represented by the formula (23):



wherein  $\text{R}_3$  has the same meaning as defined above, to form a 3-((1,2,4)-triazol-5-yl)-3-hydroxypropyl-phosphonate represented by formula (24):



wherein  $\text{R}_3$  and A have the same meanings as defined above; converting the 3-((1,2,4)-triazol-5-yl)-3-hydroxypropyl-phosphonate to 3-(1H-1,2,4-triazol-5-yl)-3-hydroxypropyl-phosphonic acid or phosphonate or oxidizing said 3-((1,2,4)-triazol-5-yl)-3-hydroxypropyl-phosphonate to form a 3-((1,2,4)-triazol-5-yl)-3-oxopropyl-phosphonate represented by formula (25)



wherein  $\text{R}_3$  and A have the same meanings as defined above, followed by conversion into 3-(1H-1,2,4-triazol-5-yl)-3-oxopropyl-phosphonic acid or phosphonate.

In the processes of this invention, A as a protective group may not be limited so long as it is conventionally used for inhibiting the reaction of triazole ring in synthetic organic chemistry and includes, for example, triphenylmethyl group, benzyl group, tert-butoxy-carbonyl group, allyl group and sulfonyl group.

Various intermediates formed in the above processes of forming the present compounds are novel compounds. Therefore the intermediate compounds represented by formulae (7), (8), (9), (10), (11), (13), (14), (15), (16), (24) and (25) are also included in this invention. Accordingly, the compounds of formula (1):



- 7 -



wherein  $\text{R}_1$  represents a hydrogen atom or a group A which is a protective group or is  $\text{C}_1$ - $\text{C}_4$ -alkyl; X and Y jointly represent a carbonyl group together with the carbon atom to which they are bonded, or X and Y each independently represent a hydrogen atom or a  $-\text{OR}_2$  group, wherein  $\text{R}_2$  represents a hydrogen atom, a lower alkyl group, an acetyl group, a benzyl group or a silyl group represented by  $-\text{SiR}'_3$  (wherein  $\text{R}'$  represents an alkyl group); and Z represents a  $-\text{CH}_2\text{PO}(\text{OR}_3)_2$  or  $-\text{CH}_2\text{OPO}(\text{OR}_3)_2$  group (wherein  $\text{R}_3$  represents a hydrogen atom, a lower alkyl group, a benzyl group, a phenyl group, a cyanoethyl group, an allyl group, a  $\text{CH}_2\text{OH}$  group, a  $\text{COOR}_5$  group, or an alkali metal, alkaline earth metal, ammonium, organic ammonium, trialkylsulphonium, trialkylsulfoxonium, phosphonium or amidinium cation; n is 0 or 1; and  $\text{R}_5$  is a  $\text{C}_1$ - $\text{C}_6$ -alkyl group, are all inclusive of the compounds of formula (2)



wherein  $\text{R}_1'$  represents a hydrogen atom or  $\text{C}_1$ - $\text{C}_4$ -alkyl; X and Y jointly represent a carbonyl group together with the carbon atom to which they are bonded, or X and Y each independently represent a hydrogen atom or a  $-\text{OR}_2$  group, wherein  $\text{R}_2$  represents a hydrogen atom, a lower alkyl group, an acetyl group, a benzyl group or a silyl group represented by  $-\text{SiR}'_3$  (wherein  $\text{R}'$  represents an alkyl group); and  $\text{Z}'$  represents a  $-\text{CH}_2\text{PO}(\text{OR}_3)_2$  or  $-\text{CH}_2\text{OPO}(\text{OR}_3)_2$  group (wherein  $\text{R}_3$  represents a hydrogen atom, a lower alkyl group, a benzyl group, a phenyl group, a cyanoethyl group, an allyl group, or an alkali metal, alkaline earth metal, ammonium, organic ammonium, trialkylsulphonium, trialkylsulfoxonium, phosphonium or amidinium cation; and n is 0 or 1; and the above intermediate compounds, and therefore constitute the gist of this invention.

- 8 -

Preferred compounds of the formula (1) are those, in which  $R_1$  represents a hydrogen atom or a group A which is a protective group; X and Y jointly represent a carbonyl group together with the carbon atom to which they are bonded, or X and Y each independently represent a hydrogen atom or a  $-OR_2$  group, wherein  $R_2$  represents a hydrogen atom, a lower alkyl group, an acetyl group, a benzyl group or a silyl group represented by  $-SiR'_3$  (wherein  $R'$  represents an alkyl group); and Z represents a  $-CH_2PO(OR_3)_2$  or  $-CH_2OPO(OR_3)_2$  group (wherein  $R_3$  represents a hydrogen atom, a lower alkyl group, a benzyl group, a phenyl group, a cyanoethyl group or an allyl group or a  $CH_2OH$  group or a  $COOR_5$  group,  $n$  is 1; and  $R_5$  is an alkyl group, preferably a  $C_1$ - $C_4$ -alkyl group.

Preferred compounds of the formula (2) are those, in which  $R_1'$  represents a hydrogen atom; X and Y jointly represent a carbonyl group together with the carbon atom to which they are bonded, or X and Y each independently represent a hydrogen atom or a  $-OR_2$  group, wherein  $R_2$  represents a hydrogen atom, a lower alkyl group, an acetyl group, a benzyl group or a silyl group represented by  $-SiR'_3$  (wherein  $R'$  represents an alkyl group); and Z' represents a  $-CH_2PO(OR_3)_2$  or  $-CH_2OPO(OR_3)_2$  group (wherein  $R_3$  represents a hydrogen atom, a lower alkyl group, a benzyl group, a phenyl group, a cyanoethyl group, an allyl group, or an alkali metal, alkaline earth metal, ammonium, organic ammonium, trialkylsulphonium, trialkylsulfoxonium, phosphonium or amidinium cation; and  $n$  is 0 or 1, preferably 1.

Most especially prominent groups of compounds of formula (2) are those wherein X and Y independently represent a  $-OR_2$  group, wherein  $R_2$  represents an acetyl group.


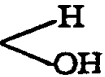

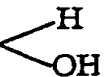

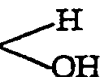
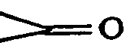
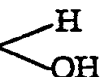
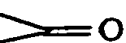
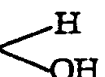

Particularly preferred of all the present compounds are described in table A:

Table A:

Compound No.	Structure
1	<chem>Oc1cncn1C(O)CCOP(=O)(O)O</chem>
2	<chem>O=C1C=CN2C(=N1)N=CN2CCOP(=O)(O)O</chem>
3	<chem>Oc1cncn1C(O)CCOP(=O)(O)O</chem>
102	<chem>Cn1cncn1C(O)CCOP(=O)(O)O</chem>
106	<chem>Oc1cncn1C(O)COP(=O)(O)O</chem>

Typical examples of these and other compounds according to this invention will be listed in the following Table 1; wherein Me represents a methyl group; Et represents an ethyl group; Bn represents a benzyl group, Acetyl represents an acetyl group; CPh<sub>3</sub> represents a triphenylmethyl group; Ph represents a phenyl group, Boc represents a 5-butoxycarbonyl group; and tBu represents a t-butyl group.

Table 1:

Comp. No.	R <sub>1</sub>		Z' *	physical data
1	H		P(O)(OH) <sub>2</sub>	m.p. 108°C
2	H		P(O)(OH) <sub>2</sub>	m.p. 196-198°C
3	H		OP(O)(OH) <sub>2</sub>	m.p. 40-43°C
4	H		OP(O)(OH) <sub>2</sub>	.
5	H		P(O)(OEt) <sub>2</sub>	oil
6	H		P(O)(OEt) <sub>2</sub>	
7	H		OP(O)(OEt) <sub>2</sub>	
8	H		OP(O)(OEt) <sub>2</sub>	
9	H		P(O)(OBn) <sub>2</sub>	
10	H		P(O)(OBn) <sub>2</sub>	

\* Z = -CH<sub>2</sub>Z'

- 11 -

Table 1 (Continuation)



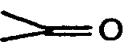



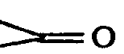




Comp. No.	R <sub>1</sub>		Z'	physical data
11	H		OP(O)(OBn) <sub>2</sub>	
12	H		OP(O)(OBn) <sub>2</sub>	
13	H		P(O)(O-CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>	
14	H		P(O)(O-CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>	
15	H		OP(O)(O-CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>	
16	H		OP(O)(O-CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>	
17	H		P(O)(O-CH <sub>2</sub> -CH=CH <sub>2</sub> ) <sub>2</sub>	
18	H		P(O)(O-CH <sub>2</sub> -CH=CH <sub>2</sub> ) <sub>2</sub>	
19	H		OP(O)(O-CH <sub>2</sub> -CH=CH <sub>2</sub> ) <sub>2</sub>	
20	H		OP(O)(O-CH <sub>2</sub> -CH=CH <sub>2</sub> ) <sub>2</sub>	

Table 1 (Continuation)


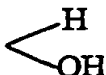

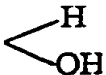
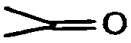
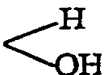
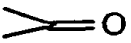
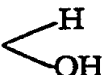

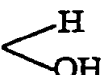


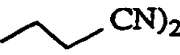
Comp. No.	R <sub>1</sub>		Z'	physical data
21	CPh <sub>3</sub>		P(O)(OEt) <sub>2</sub>	m.p. 190-192°C
22	CPh <sub>3</sub>		P(O)(OEt) <sub>2</sub>	
23	CPh <sub>3</sub>		OP(O)(OEt) <sub>2</sub>	
24	CPh <sub>3</sub>		OP(O)(OEt) <sub>2</sub>	
25	CPh <sub>3</sub>		P(O)(OBn) <sub>2</sub>	
26	CPh <sub>3</sub>		P(O)(OBn) <sub>2</sub>	
27	CPh <sub>3</sub>		OP(O)(OBn) <sub>2</sub>	
28	CPh <sub>3</sub>		OP(O)(OBn) <sub>2</sub>	
29	CPh <sub>3</sub>		P(O)(O-  ) <sub>2</sub>	
30	CPh <sub>3</sub>		P(O)(O-  ) <sub>2</sub>	

Table 1 (Continuation)












Comp. No.	R <sub>1</sub>		Z'	physical data
31	CPh <sub>3</sub>		OP(O)(O-CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>	
32	CPh <sub>3</sub>		OP(O)(O-CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>	
33	CPh <sub>3</sub>		P(O)(O-CH <sub>2</sub> -CH=CH <sub>2</sub> ) <sub>2</sub>	
34	CPh <sub>3</sub>		P(O)(O-CH <sub>2</sub> -CH=CH <sub>2</sub> ) <sub>2</sub>	
35	CPh <sub>3</sub>		OP(O)(O-CH <sub>2</sub> -CH=CH <sub>2</sub> ) <sub>2</sub>	
36	CPh <sub>3</sub>		OP(O)(O-CH <sub>2</sub> -CH=CH <sub>2</sub> ) <sub>2</sub>	
37	CPh <sub>3</sub>		P(O)(OEt) <sub>2</sub>	m.p. 165-166°C
38	CPh <sub>3</sub>		OP(O)(OEt) <sub>2</sub>	
39	CPh <sub>3</sub>		P(O)(OBn) <sub>2</sub>	
40	CPh <sub>3</sub>		OP(O)(OBn) <sub>2</sub>	

Table 1 (Continuation)

Comp. No.	R <sub>1</sub>	$\begin{array}{c} \diagup \text{X} \\ \diagdown \text{Y} \end{array}$	Z'	physical data
41	CPh <sub>3</sub>	$\begin{array}{c} \text{H} \\ \diagdown \text{OAcetyl} \end{array}$	P(O)(O-CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>	
42	CPh <sub>3</sub>	$\begin{array}{c} \text{H} \\ \diagdown \text{OAcetyl} \end{array}$	OP(O)(O-CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>	
43	CPh <sub>3</sub>	$\begin{array}{c} \text{H} \\ \diagdown \text{OAcetyl} \end{array}$	P(O)(O-CH=CH) <sub>2</sub>	
44	CPh <sub>3</sub>	$\begin{array}{c} \text{H} \\ \diagdown \text{OAcetyl} \end{array}$	OP(O)(O-CH=CH) <sub>2</sub>	
45	CPh <sub>3</sub>	$\begin{array}{c} \text{H} \\ \diagdown \text{OSiR}_3 \end{array}$	P(O)(OEt) <sub>2</sub>	
46	CPh <sub>3</sub>	$\begin{array}{c} \text{H} \\ \diagdown \text{OSiR}_3 \end{array}$	OP(O)(OEt) <sub>2</sub>	
47	CPh <sub>3</sub>	$\begin{array}{c} \text{H} \\ \diagdown \text{OSiR}_3 \end{array}$	P(O)(OBn) <sub>2</sub>	
48	CPh <sub>3</sub>	$\begin{array}{c} \text{H} \\ \diagdown \text{OSiR}_3 \end{array}$	OP(O)(OBn) <sub>2</sub>	
49	CPh <sub>3</sub>	$\begin{array}{c} \text{H} \\ \diagdown \text{OSiR}_3 \end{array}$	P(O)(O-CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>	
50	CPh <sub>3</sub>	$\begin{array}{c} \text{H} \\ \diagdown \text{OSiR}_3 \end{array}$	OP(O)(O-CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>	














- 15 -

Table 1 (Continuation)

Comp. No.	R <sub>1</sub>	$\begin{array}{c} \diagup X \\ \diagdown Y \end{array}$	Z'	physical data
51	CPh <sub>3</sub>	$\begin{array}{c} \diagup H \\ \diagdown OSiR_3 \end{array}$	P(O)(O-CH <sub>2</sub> -CH=CH <sub>2</sub> ) <sub>2</sub>	
52	CPh <sub>3</sub>	$\begin{array}{c} \diagup H \\ \diagdown OSiR_3 \end{array}$	OP(O)(O-CH <sub>2</sub> -CH=CH <sub>2</sub> ) <sub>2</sub>	
53	CPh <sub>3</sub>	$\begin{array}{c} \diagup H \\ \diagdown OBn \end{array}$	P(O)(OEt) <sub>2</sub>	
54	CPh <sub>3</sub>	$\begin{array}{c} \diagup H \\ \diagdown OSiR_3 \end{array}$	OP(O)(OEt) <sub>2</sub>	
55	CPh <sub>3</sub>	$\begin{array}{c} \diagup H \\ \diagdown OSiR_3 \end{array}$	P(O)(OBn) <sub>2</sub>	
56	CPh <sub>3</sub>	$\begin{array}{c} \diagup H \\ \diagdown OSiR_3 \end{array}$	OP(O)(OBn) <sub>2</sub>	
57	CPh <sub>3</sub>	$\begin{array}{c} \diagup H \\ \diagdown OSiR_3 \end{array}$	P(O)(O-CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>	
58	CPh <sub>3</sub>	$\begin{array}{c} \diagup H \\ \diagdown OSiR_3 \end{array}$	OP(O)(O-CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>	
59	CPh <sub>3</sub>	$\begin{array}{c} \diagup H \\ \diagdown OSiR_3 \end{array}$	P(O)(O-CH <sub>2</sub> -CH=CH <sub>2</sub> ) <sub>2</sub>	
60	CPh <sub>3</sub>	$\begin{array}{c} \diagup H \\ \diagdown OSiR_3 \end{array}$	OP(O)(O-CH <sub>2</sub> -CH=CH <sub>2</sub> ) <sub>2</sub>	

- 16 -

Table 1 (Continuation)

Comp. No.	R <sub>1</sub>		Z'	physical data
61	Bn		P(O)(OH) <sub>2</sub>	
62	Bn		P(O)(OH) <sub>2</sub>	
63	Bn		OP(O)(OH) <sub>2</sub>	
64	Bn		OP(O)(OH) <sub>2</sub>	
65	Bn		P(O)(OEt) <sub>2</sub>	
66	Bn		P(O)(OEt) <sub>2</sub>	
67	Bn		OP(O)(OEt) <sub>2</sub>	
68	Bn		OP(O)(OEt) <sub>2</sub>	
69	Bn		P(O)(OBn) <sub>2</sub>	
70	Bn		P(O)(OBn) <sub>2</sub>	

- 17 -

Table 1 (Continuation)












Comp. No.	R <sub>1</sub>		Z'	physical data
71	Bn		OP(O)(OBn) <sub>2</sub>	
72	Bn		OP(O)(OBn) <sub>2</sub>	
73	Bn		P(O)(O-CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>	
74	Bn		P(O)(O-CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>	
75	Bn		OP(O)(O-CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>	
76			OP(O)(O-CH <sub>2</sub> -CH <sub>2</sub> -CN) <sub>2</sub>	
77	Bn		P(O)(O-CH <sub>2</sub> -CH=CH <sub>2</sub> ) <sub>2</sub>	
78	Bn		P(O)(O-CH <sub>2</sub> -CH=CH <sub>2</sub> ) <sub>2</sub>	
79	Bn		OP(O)(O-CH <sub>2</sub> -CH=CH <sub>2</sub> ) <sub>2</sub>	
80	Bn		OP(O)(O-CH <sub>2</sub> -CH=CH <sub>2</sub> ) <sub>2</sub>	

Table 1 (Continuation)











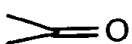















Comp. No.	R <sub>1</sub>		Z'	physical data
81	CPh <sub>3</sub>		P(O)(OtBu) <sub>2</sub>	
82	CPh <sub>3</sub>		P(O)(OtBu) <sub>2</sub>	
83	CPh <sub>3</sub>		OP(O)(OtBu) <sub>2</sub>	
84	CPh <sub>3</sub>		OP(O)(OtBu) <sub>2</sub>	
85	Bn		P(O)(OtBu) <sub>2</sub>	
86	Bn		P(O)(OtBu) <sub>2</sub>	
87	Bn		OP(O)(OtBu) <sub>2</sub>	
88	Bn		OP(O)(OtBu) <sub>2</sub>	
89	Boc		P(O)(OtBu) <sub>2</sub>	
90	Boc		P(O)(OtBu) <sub>2</sub>	

Table 1 (Continuation)

Comp. No.	R <sub>1</sub>		Z'	physical data
91	Boc		OP(O)(OtBu) <sub>2</sub>	
92	Boc		OP(O)(OtBu) <sub>2</sub>	
93	CPh <sub>3</sub>		P(O)(OtBu) <sub>2</sub>	
94	Bn		P(O)(OtBu) <sub>2</sub>	
95	Boc		P(O)(OtBu) <sub>2</sub>	
96	CPh <sub>3</sub>		OP(O)(OtBu) <sub>2</sub>	
97	Bn		OP(O)(OtBu) <sub>2</sub>	
98	Boc		OP(O)(OtBu) <sub>2</sub>	
99	CPh <sub>3</sub>		OP(O)(OtBu) <sub>2</sub>	
100	Bn		OP(O)(OtBu) <sub>2</sub>	
101	CH <sub>3</sub>		P(O)(OC <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>	oil
102	CH <sub>3</sub>		P(O)(OH) <sub>2</sub>	m.p. 64-78°C
103	C <sub>2</sub> H <sub>5</sub>		P(O)(OH) <sub>2</sub>	m.p. 127-130°C
104	C <sub>2</sub> H <sub>5</sub>		P(O)(OH) <sub>2</sub>	

- 20 -

Comp. No.	R <sub>1</sub>	$\begin{array}{c} \diagup X \\ \diagdown Y \end{array}$	Z'	physical data
105	H	$\begin{array}{c} \diagup H \\ \diagdown OAcetyl \end{array}$	P(O)(OH) <sub>2</sub>	m.p. 100°C
106	H	$\begin{array}{c} \diagup H \\ \diagdown OH \end{array}$	OP(O)(OH) <sub>2</sub>	

The compounds of formula (1), in particular represented by formula (2), are distinguished by microbicidal and herbicidal properties which render them excellent for use in crops of useful plants, in particular in cereals, cotton, soya, rape, maize and rice.

The invention also relates to herbicidal and microbicidal compositions which comprise a novel active ingredient of formula (2), and methods for inhibition of plant growth.

The active ingredients of formula (2) are as a rule employed successfully at rates of application of 0.001 to 4 kg/ha, in particular 0.005 to 2 kg/ha. The doses required for the desired action can be determined by experiments. It depends on the nature of the action, the development stage of the crop plant and of the weed and on the application conditions (location, time, method) and can, as a result of these parameters, be varied within wide ranges.

The compounds of formula (2) are employed in unaltered form, as obtainable by the synthesis, or preferably together with the auxiliaries conventionally used in formulation technology, and they are therefore processed in a known manner to give, for example, emulsifiable concentrates, directly sprayable or dilutable solutions, dilute emulsions, wettable powders, soluble powders, dusts, granules, and also encapsulations, for example in polymeric substances. The application methods, such as spraying, atomising, dusting, scattering or pouring, as well as the type of compositions are selected to suit the intended aims and the prevailing circumstances.

The formulations, i.e. the compositions, preparations or combinations comprising the active substance of formula (2) and, if desired, one or more solid or liquid additives, are

prepared in a known manner, for example by intimately mixing and/or grinding the active substances with extenders, for example with solvents, solid carriers and, if desired, surface-active compounds (surfactants).

The following are possible as solvents: aromatic hydrocarbons, in particular the fractions  $C_8$  to  $C_{12}$ , such as mixtures of alkylbenzenes, for example xylene mixtures or alkylated naphthalenes; aliphatic and cycloaliphatic hydrocarbons such as paraffins, cyclohexane or tetrahydronaphthalene; alcohols, such as ethanol, propanol or butanol; glycols as well as their ethers and esters, such as propylene glycol or dipropylene glycol ether, ketones such as cyclohexanone, isophorone or diacetone alcohol, strongly polar solvents such as N-methyl-2-pyrrolidone, dimethyl sulfoxide or water; vegetable oils as well as their esters, such as rapeseed oil, castor oil or soybean oil; and if appropriate also silicone oils.

Suitable surface-active compounds are non-ionic, cationic and/or anionic surfactants having good emulsifying, dispersing and wetting properties, depending on the nature of the active substance of formula (2) to be formulated. Surfactants are also to be understood as meaning mixtures of surfactants.

Anionic surfactants which are suitable can be either so-called water-soluble soaps or water-soluble synthetic surface-active compounds.

Suitable soaps which may be mentioned are the alkali metal salts, alkaline earth metal salts or substituted or unsubstituted ammonium salts of higher fatty acids ( $C_{10}$ - $C_{22}$ ), such as the Na salts or K salts of oleic or stearic acid, or of natural mixtures of fatty acids which can be obtained, for example, from coconut oil or tallow oil. Mention must also be made of the fatty acid methyltaurates.

However, so-called synthetic surfactants are used more frequently, in particular fatty alcohol sulfonates, fatty alcohol sulfates, sulfonated benzimidazole derivatives or alkylarylsulfonates.

The fatty alcohol sulfonates or fatty alcohol sulfates are generally in the form of alkali metal salts, alkaline earth metal salts or substituted or unsubstituted ammonium salts, and have an alkyl radical having 8 to 22 C atoms, alkyl also including the alkyl moiety of acyl radicals, for example the Na or Ca salt of ligninsulfonic acid, of the dodecylsulfuric ester or of a fatty alcohol sulfate mixture prepared from natural fatty acids. This group also includes the salts of the sulfuric esters and sulfonic acids of fatty alcohol/ethylene oxide

adducts. The sulfonated benzimidazole derivatives preferably contain 2 sulfonyl groups and one fatty acid radical having 8 to 22 C atoms. Examples of alkylarylsulfonates are the Na, Ca or triethanolamine salts of dodecylbenzenesulfonic acid, of dibutyl-naphthalenesulfonic acid or of a naphthalenesulfonic acid/formaldehyde condensation product.

Other suitable compounds are the corresponding phosphates, such as the salts of the phosphoric ester of a p-nonylphenol/(4-14)-ethylene oxide adduct, or phospholipids.

Suitable non-ionic surfactants are mainly polyglycol ether derivatives of aliphatic or cycloaliphatic alcohols, of saturated or unsaturated fatty acids and of alkylphenols, which can contain 3 to 30 glycol ether groups and 8 to 20 carbon atoms in the (aliphatic) hydrocarbon radical and 6 to 18 carbon atoms in the alkyl radical of the alkylphenols.

Other non-ionic surfactants which are suitable are the water-soluble polyethylene oxide adducts with polypropylene glycol, ethylenediaminopolypropylene glycol and alkylpolypropylene glycol which have 1 to 10 carbon atoms in the alkyl chain and which contain 20 to 250 ethylene glycol ether groups and 10 to 100 propylene glycol ether groups. The abovementioned compounds customarily contain 1 to 5 ethylene glycol units per propylene glycol unit.

Examples of non-ionic surfactants which may be mentioned are nonylphenolpolyethoxyethanols, castor oil polyglycol ethers, polypropylene/polyethylene oxide adducts, tributylphenoxypolyethoxyethanol, polyethylene glycol and octylphenoxypolyethoxyethanol.

Other suitable substances are fatty acid esters of polyoxyethylenesorbitan, such as polyoxyethylenesorbitan trioleate.

The cationic surfactants are mainly quaternary ammonium salts, which contain at least one alkyl radical having 8 to 22 C atoms as N-substituents and which have lower halogenated or free alkyl, benzyl or lower hydroxyalkyl radicals as further substituents. The salts are preferably in the form of halides, methylsulfates or ethylsulfates, for example stearyltrimethylammonium chloride or benzyldi(2-chloroethyl)ethylammonium bromide.

The surfactants customary in formulation technology are described, inter alia, in the



following publications:

"McCutcheon's Detergents and Emulsifiers Annual", Mc Publishing Corp., Glen Rock, New Jersey, 1988;

M. and J. Ash. "Encyclopedia of Surfactants", Vol. I-III, Chemical Publishing Co., New York, 1980-1981.

Dr. Helmut Stache, "Tensid-Taschenbuch [Surfactant Guide]", Carl Hanser Verlag, Munich, Vienna, 1981;

As a rule, the pesticidal preparations contain 0.1 to 99 %, in particular 0.1 to 95 %, of the active substance of formula (2), 1 to 99 % of a solid or liquid additive and 0 to 25 %, in particular 0.1 to 25 %, of a surfactant.

While concentrated compositions are more preferred as commercial goods, the user generally uses dilute compositions.

The compositions can also comprise further additives such as stabilisers, for example epoxidised or unepoxidised vegetable oils (epoxidised coconut oil, rapeseed oil or soybean oil), defoamers, for example silicone oil, preservatives, viscosity regulators, binders, tackifiers, as well as fertilisers or other active substances for achieving specific effects.

In particular, preferred formulations have the following composition: (% = per cent by weight)

Emulsifiable concentrates:

Active ingredient:	1 to 90 %, preferably 5 to 20 %
Surface-active agent:	1 to 30 %, preferably 10 to 20 %
Liquid carrier:	50 to 94 %, preferably 70 to 85 %

Suspension concentrates:

Active ingredient:	5 to 75 %, preferably 10 to 50 %
Water:	94 to 24 %, preferably 88 to 30 %
Surface-active agent:	1 to 40 %, preferably 2 to 30 %

As a rule, the active substances of formula (2) are successfully employed at application

- 24 -

rates from 0.001 to 10 kg/ha, in particular 0.005 to 2 kg/ha. The dosage rate which is required for the desired action can be determined by tests. It depends on the nature of the action, the development stage of the crop plant and the weed, as well as on the application (location, time, method) and, due to these parameters, can vary within wide limits.

#### Controlled release of active substance

The dissolved active substance is applied to mineral granule carriers or polymerised granules (urea/formaldehyde) and allowed to dry. If desired, a coating can be applied (coated granules), which permits slow release of the active substance over a certain period.

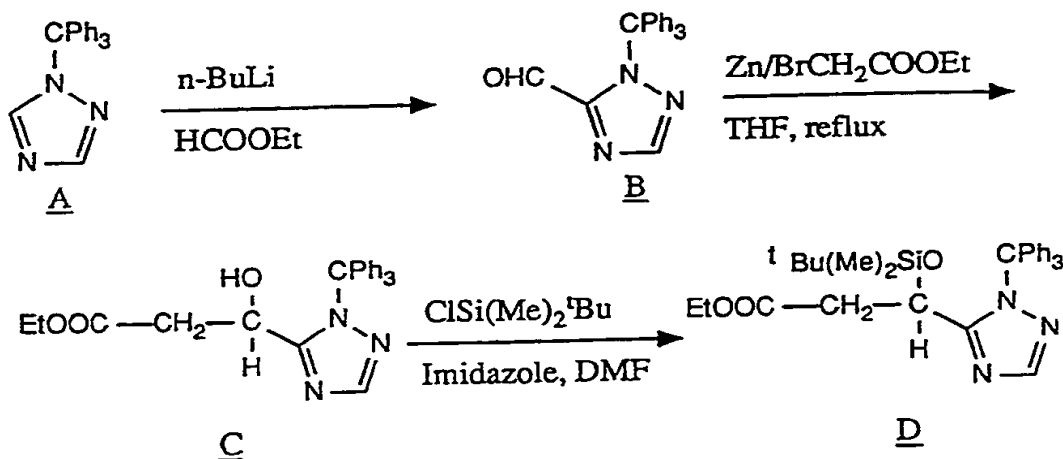
This invention further relates to the use of at least one of the compounds represented by formula (2) as the active ingredient for inhibiting enzymes which participate in the biosynthesis of histidine.

#### Example

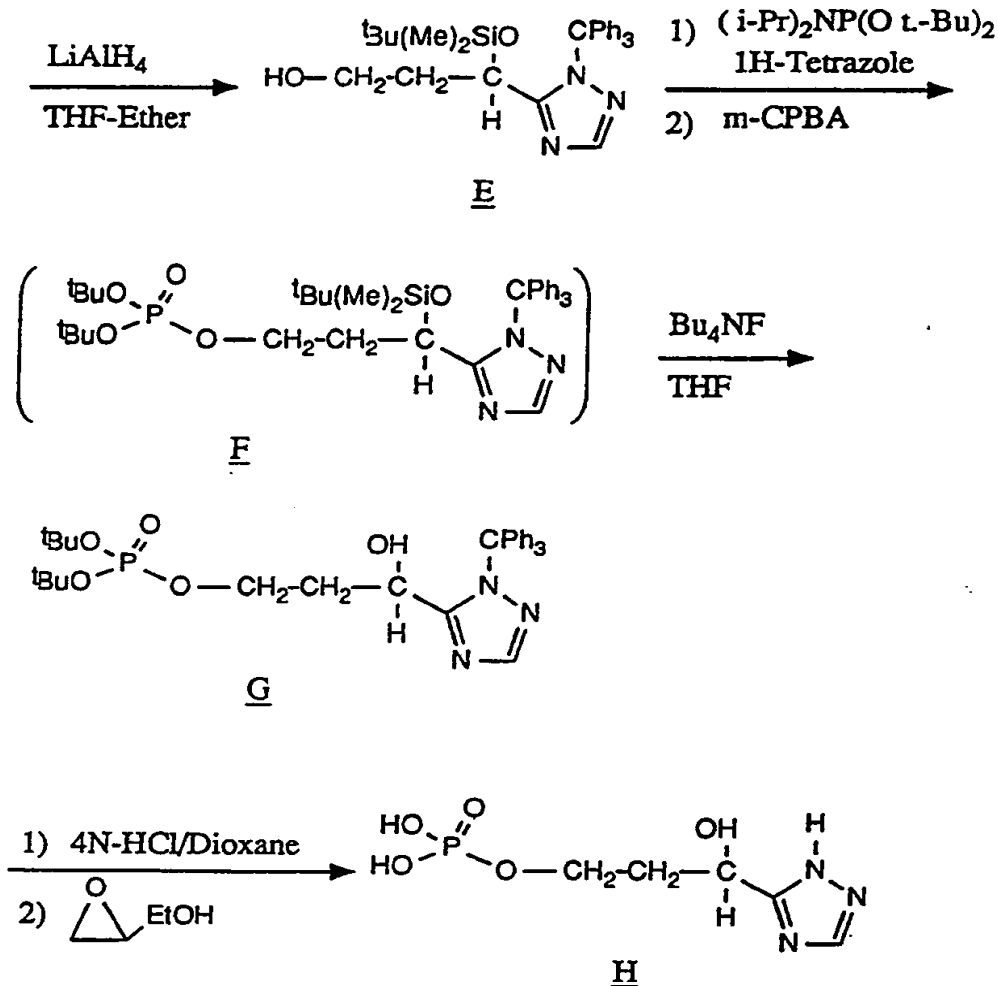
This invention will be described below more specifically by way of Examples.

#### Example 1

A phosphate compound of this invention was prepared according to the flow charts shown below:



- 25 -



(1) ((1-Triphenylmethyl)-(1,2,4)-triazol-5-yl)aldehyde: Compound B

To a stirred solution of 1-triphenylmethyl-(1,2,4)-triazole (Compound A) (40 g, 0.129 mole) dissolved in 400 ml of THF was added n-BuLi (1.5 M hexane solution, 112 ml) at  $-78^\circ\text{C}$ , and after the resulting mixture was stirred at the same temperature for 20 minutes, 17 g (0.23 mole) of ethyl formate was added thereto. After the resulting mixture was stirred for 10 minutes, the temperature of the reaction mixture was elevated to room temperature and then quenched to  $-78^\circ\text{C}$ , followed by addition of a saturated aqueous  $\text{NH}_4\text{Cl}$  and extraction with  $\text{CHCl}_3$ . The  $\text{CHCl}_3$  extract was collected, washed with brine, dried over  $\text{MgSO}_4$  and concentrated. The residue was purified by recrystallization from ether to give 40 g of the desired compound B as a colorless powder (Yield 92 %).

$^1\text{H-NMR}$  ( $\text{CDCl}_3$ ,  $\delta$ ): 7.04-7.40 (15H, m) 8.09 (1H, s), 9.14 (1H, s)

- 26 -

IR (CHCl<sub>3</sub>, cm<sup>-1</sup>): 3075, 3025, 1720, 1708, 1600, 1495, 1450, 1430, 1335, 1325, 1285, 1220, 1100, 1075, 1040, 905, 890, 785, 760, 700, 670

(2) Ethyl 3-hydroxy-3-(1-triphenylmethyl-(1,2,4)-triazol-5-yl)-propionate: Compound C

To a stirred solution of Compound B (10 g, 29.5 mmol) dissolved in 200 ml of THF were added successively 3.86 g (59 mmol) of a zinc powder and 6.54 ml (59 mmol) of ethyl acetate, and after the resulting mixture was refluxed for one hour, it was filtered through Celite. The filtrate was concentrated to give an oily residue, which was purified over silica gel column chromatography (hexane/ethyl acetate = 3:1 - 1:2) to obtain 5.84 g of Compound C as a substantially colorless oil (Yield 46.4 %).

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, δ): 1.10 (3H, t, J=7.0 Hz), 2.97 (2H, d, J=7.9 Hz), 3.95 (2H, q, J=7.0 Hz), 4.22 (1H, brs), 6.35 (1H, t, J=7.0 Hz), 6.8-7.5 (15H, m), 8.67 (1H, s)

IR (neat, cm<sup>-1</sup>): 3430, 3150, 3070, 3000, 1730, 1600, 1530, 1490, 1470, 1450, 1400, 1375, 1330, 1270, 1220, 1190, 1165, 1090, 1065, 1040, 1005, 880, 760, 770

(3) Ethyl 3-tert-butyldimethylsilyloxy-3-(1-triphenylmethyl-(1,2,4)-triazol-5-yl)-propionate: Compound D

To a stirred solution of Compound C (12 g, 28.1 mmol) dissolved in 100 ml of DMF were added successively 6.34 g (42 mmol) of tert-butyldimethylsilyl chloride and 3.82 g (56.2 mmol) of imidazole, and after the mixture was stirred at room temperature for 3 hours, the reaction mixture was quenched with ice water, extracted with ethyl acetate, washed with brine, dried over MgSO<sub>4</sub> and concentrated. The residue was purified over silica gel column chromatography (hexane/ethyl acetate = 4:1) to obtain 14.17 g of Compound D as a colorless oil (Yield 93.2 %).

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, δ): 0.00 (3H, s), 0.13 (3H, s), 0.92 (9H, s), 1.33 (3H, t, J=7.3 Hz), 3.0 (2H, m), 4.17 (2H, q, J=7.3 Hz), 5.45 (1H, dd, J=5.5 Hz, J=8 Hz), 7.2-7.47 (15H, m), 8.0 (1H, s)

IR (neat, cm<sup>-1</sup>): 2950, 2930, 2855, 1735, 1595, 1490, 1470, 1445, 1370, 1250, 1170, 1100, 1030, 1000, 955, 880, 840, 780, 760, 700

(4) 3-tert-Butyldimethylsilyloxy-3-(1-triphenylmethyl-(1,2,4)-triazol-5-yl)propanol: Compound E

To a stirred solution of Compound D (4.8 g, 8.9 mmol) dissolved in THF/ether (60 ml:

- 27 -

30 ml) was added 505 mg (13.3 mmoles) of  $\text{LiAlH}_4$  at  $-10^\circ\text{C}$ , and after the mixture was stirred for 1 hour, the reaction mixture was quenched with 1N-KOH solution and then filtered through Celite. The filtrate was concentrated to give an oily residue, which was purified over silica gel chromatography (hexane/ethyl acetate = 1:1) to obtain 2.92 g of the desired compound E as a colorless oil (Yield 66 %).

$^1\text{H-NMR}$  ( $\text{CDCl}_3$ ,  $\delta$ ): -0.13 (3H, s), 0.00 (3H, s), 0.92 (9H, s), 2.20 (2H, M), 2.70 (1H, brs), 3.88 (2H, m), 5.20 (1H, t  $J=6.3$  Hz), 7.15-7.45 (15H, m), 8.0 (1H, s)  
IR (neat,  $\text{cm}^{-1}$ ): 3375, 3060, 3030, 2950, 2850, 2880, 1600, 1510, 1490, 1450, 1390, 1360, 1325, 1250, 1190, 1170, 1000, 1030, 985, 940, 905, 880, 840, 770, 700, 670, 640

(5) Di-tert-butyl 3-hydroxy-3-(1-triphenylmethyl-(1,2,4)-triazol-5-yl)propyl-phosphate:  
Compound G

To a stirred solution of Compound E (7 g, 14 mmoles) dissolved in THF/ $\text{CH}_3\text{CN}$  (40 ml : 40 ml) were successively added 11.7 g (42.2 mmoles) of bis(tert-butoxy)(diisopropyl-amino)phosphine and 2.95 g (42 mmoles) of 1H-tetrazole at room temperature, and after the mixture was stirred for 1 hour, the reaction mixture was cooled to  $-40^\circ\text{C}$ . To the mixture was slowly added a solution of m-CPBA (7.26 g, 42 mmoles) dissolved in  $\text{CH}_2\text{Cl}_2$  (50 ml), and the resulting mixture was stirred at the same temperature for 10 minutes. After the temperature of the reaction mixture was elevated to room temperature, it was extracted with  $\text{CH}_2\text{Cl}_2$ . The extract was collected, washed successively with a 5 % aqueous  $\text{NaHSO}_3$  solution, a saturated aqueous  $\text{NaHCO}_3$  and brine, dried over  $\text{MgSO}_4$  and concentrated to give an oily residue (Compound F), and then THF (100 ml) and TBAF (1M THF solution, 40 ml) were added thereto at 0 to  $5^\circ\text{C}$ .

After the resulting mixture was stirred at room temperature for 18 hours, the reaction mixture was extracted with  $\text{CHCl}_3$ . The extract was collected, washed with water and brine, dried over  $\text{MgSO}_4$  and concentrated to give an oily residue.

The thus obtained product was purified over silica gel chromatography ( $\text{CHCl}_3$ /acetone = 4:1) to obtain 5 g of Compound G as a colorless oil (Yield 62 %).

$^1\text{H-NMR}$  ( $\text{CDCl}_3$ ,  $\delta$ ): 1.47 (18H, s), 2.23 (2H, m), 3.56 (1H, brs), 4.18 (2H, m), 5.05 (1H, dd  $J=5.0$  Hz,  $J=8.0$  Hz), 7.05-7.4 (15H, m), 7.91 (1H, s)  
IR (neat,  $\text{cm}^{-1}$ ): 3350, 3070, 3025, 2975, 2940, 2910, 2875, 1720, 1670, 1600, 1510, 1495,

- 28 -

1478, 1450, 1395, 1370, 1345, 1330, 1250, 1170, 1100, 1020, 920, 880, 830, 800, 750, 705, 670, 640

(6) 3-Hydroxy-3-(1H-(1,2,4)-triazol-5-yl)propyl-phosphate: Compound H

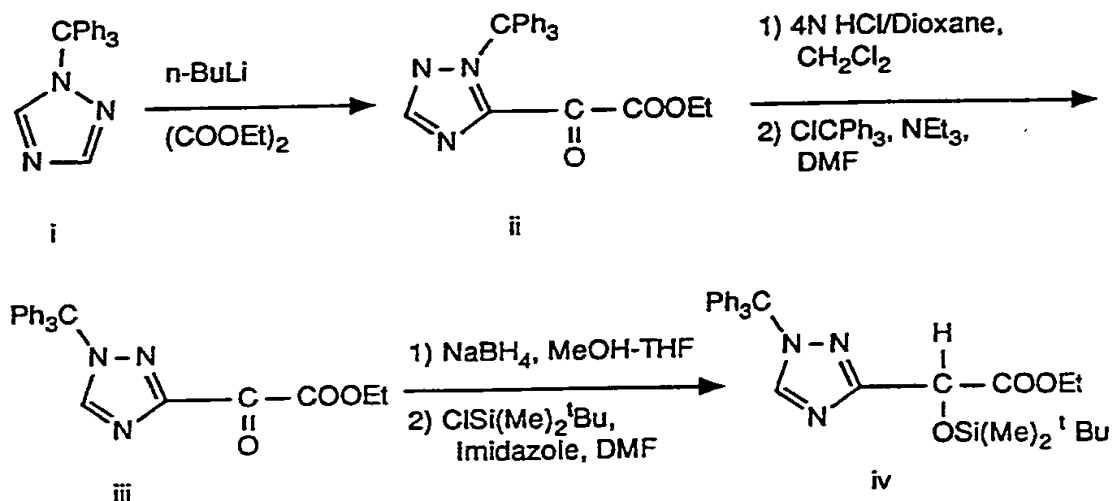
To a stirred solution of Compound G (1.0 g, 1.74 mmol) dissolved in  $\text{CH}_2\text{Cl}_2$  (5 ml) was added 8 ml of 4N-HCl-dioxane solution at 0 to 5°C, and the resulting mixture was stirred at room temperature for 1 hour. After the solvent was removed, the oily residue was diluted with 4 ml of EtOH, and then 3 ml of propylene oxide and ether were added thereto. Crystals thus precipitated were collected, washed with ether and acetone and dried in vacuo to give 290 mg of Compound H as a colorless powder (Yield 74.6 %).

$^1\text{H-NMR}$  ( $\text{D}_2\text{O}$ ,  $\delta$ ): 2.25 (2H, m), 4.05 (2H, m), 5.30 (1H, t,  $J=6.0$  Hz), 9.04 (1H, s)  
m.p. 40 to 43°C

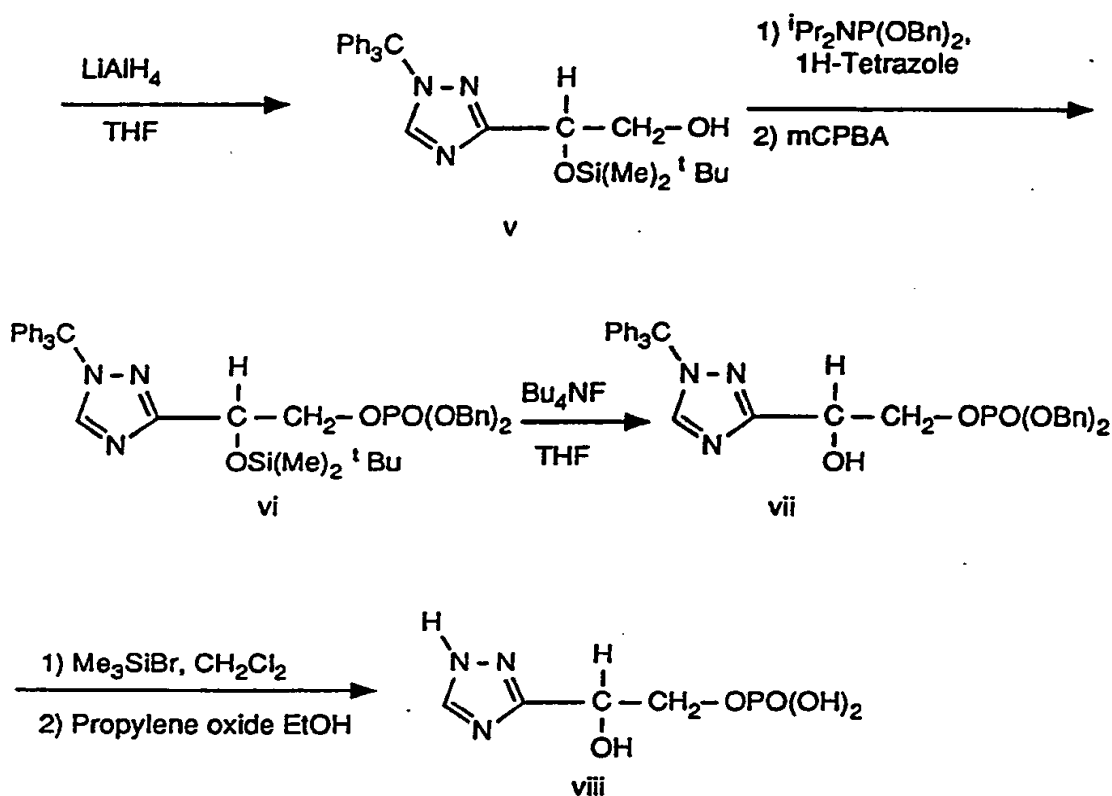
The thus obtained compound corresponds to the substance of Compound No. 3 in table 1.

Example 2

A phosphate compound of this invention was prepared according to the flow charts shown below:



- 29 -



(1) Ethyl 2-((1,2,4)-triazol-5-yl)-2-oxoacetate: Compound (ii)

To a stirred solution of 1-triphenylmethyl-(1,2,4)-triazole (compounds i) (50 g, 0.16 mol) dissolved in 750 ml of THF was added n-BuLi (1.5 M hexane solution, 0.24 mol) at  $-78^\circ\text{C}$ . The resulting mixture was stirred for 30 min at the same temperature and 42 ml (0.31 mol) of diethyl oxalate was added thereto. After stirring for 10 min, the reaction mixture was warmed to room temperature and then quenched with saturated aqueous  $\text{NH}_4\text{Cl}$ , extracted with  $\text{CHCl}_3$ . The organic layer was collected, washed with brine, dried over  $\text{MgSO}_4$  and concentrated in vacuo. The residue was purified by recrystallization from ether to give 26 g of the compound (ii) as a colorless powder (Yield 39 %).

$^1\text{H-NMR}$  ( $\text{CDCl}_3, \delta$ ): 8.07 (1H,s), 7.1-7.7 (15H,m) 4.27 (2H,q,J=7.0 Hz), 1.30 (3H,t, J=7.0 Hz)

(2) Ethyl 2-(1,2,4)-triazol-3-yl)-2-oxoacetate: Compound (iii)

To a stirred solution of compound (ii) dissolved in 150 ml of  $\text{CH}_2\text{Cl}_2$  was added 35 ml

- 30 -

(140 mmol) of 4N HCl-dioxane at room temperature. After stirring for 30 min, the solvent was evaporated and the residue was dissolved in 50 ml of DMF. To this solution were added successively 1.8 g (6.5 mmol) of triphenylmethyl chloride and 23 ml (165.3 mmol) of triethylamine at 0-5°C. After the mixture was stirred to 1 h at room temperature, the reaction mixture was quenched with ice water extracted with ethyl acetate, washed with brine, dried over  $\text{MgSO}_4$  and concentrated. The residue was purified by recrystallization from ether to give 18.8 g of the compound (iii) as a colorless powder (Yield 72,3 %).

$^1\text{H-NMR}$  ( $\text{CDCl}_3$ ,  $\delta$ ): 8.09 (1H,s), 7.0-7.6 (15H,m), 4.40 (2H,q,J=7.0 Hz), 1.30 (3H,t,J=7.0 Hz) IR (NaCl,  $\text{cm}^{-1}$ ): 1750, 1710, 1690, 1490, 1465, 1440, 1255, 1170, 1090, 1050, 1035, 1010, 970, 870, 750, 700

(3) Ethyl 2-tert-butyltrimethylsilyloxy-2-(1-triphenylmethyl-(1,2,4)-triazol-3-yl)-acetate:  
Compound (iv)

To a stirred suspension of  $\text{NaBH}_4$  (556 mg, 14,87 mmol) in MeOH was added 5 g of compound (iii) (12.15 mmol) dissolved in MeOH-THF (15 ml-35 ml) at -40°C. After the mixture was stirred for 1 h, the reaction mixture was quenched with ice-water, extracted with  $\text{CHCl}_3$ , washed with brine, dried over  $\text{MgSO}_4$  and concentrated in vacuo. The residue was dissolved in 60 ml of DMF and then 2.0 g (13.27 mmol) of tert-butyltrimethylsilyl chloride and 1.2 g (17.64 mmol) of imidazole were successively added thereto at room temperature. After the mixture was stirred for 3 h, the reaction mixture was quenched with ice-water, extracted with  $\text{CHCl}_3$ , washed with brine, dried over  $\text{MgSO}_4$  and concentrated in vacuo. The residue was purified over silica gel column chromatography ( $\text{CHCl}_3$ :acetone = 10:1) to give 4.36 g of the compound (iv) as colorless oil (Yield 74.2 %)

$^1\text{H-NMR}$  ( $\text{CDCl}_3$ ,  $\delta$ ): 7.91 (1H,s), 7.0-7.50 (15H,m), 5.43 (1H,s), 4.22 (2H,q,J=5.6 Hz), 1.25 (3H,t,J=5.6 Hz), 0.87 (9H,s), 0.10 (3H,s), 0.03 (3H,s)  
IR (neat,  $\text{cm}^{-1}$ ): 3060, 2955, 2925, 2900, 2860, 1760, 1738, 1595, 1490, 1470, 1460, 1445, 1370, 1360, 1340, 1330, 1250, 1130, 1030, 1000, 940, 880, 840, 780, 750, 700

(4) 2-tert-Butyltrimethylsilyloxy-2-(1-triphenylmethyl-(1,2,4)-triazol-3-yl)ethanol:  
Compound (v)

To a stirred solution of compound (iv) (10 g, 18.95 mmol) dissolved in 200 ml of THF was added 270 mg (7.11 mmol) of  $\text{LiAlH}_4$  at -15°C. After the mixture was stirred for 10 min, the reaction mixture was quenched with ice-water and then filtered through a



- 31 -

column of celite. The filtrate was concentrated to give oily residue which was purified over silica gel column chromatography (hexane:AcOEt = 2:1) to give 2.8 g of the compound (v) as a colorless powder (Yield 30.4 %).

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, δ): 7.98 (1H,s), 7.0-7.6 (15H,m), 4.97 (1H,m), 3.95 (2H,m), 3.30 (1H,brs), 0.91 (9H,s), 0.09 (3H,s), 0.0 (3H,s)  
IR (NaCl, cm<sup>-1</sup>): 3350, 2925, 2850, 1600, 1505, 1490, 1470, 1460, 1445, 1355, 1250, 1170, 1100, 1060, 1040, 955, 870, 840, 780, 750, 700

(5) Dibenzyl 2-tert-butyldimethylsilyloxy-2-(triphenylmethyl-(1,2,4)-triazol-3-yl)ethyl phosphate: Compound (vi)

To a stirred solution of compound (v) 2 g, 4.12 mmol) dissolved in 60 ml of CH<sub>3</sub>CN were successively added 2.65 g (7.67 mmol) of bis(benzyloxy)-(diisopropylamino)phosphine and 869 mg (12.4 mmol) of 1H-tetrazole at room temperature and the mixture was stirred for 1 h. A solution of m-CPBA (1.53 g, 8.86 mmol) dissolved in 50 ml of CH<sub>2</sub>Cl<sub>2</sub> was added slowly at -78°C and the mixture was warmed to room temperature. The reaction was quenched with ice-water and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined extracts were washed successively with 5 % aqueous NaHSO<sub>3</sub>, saturated aqueous NaHCO<sub>3</sub>, brine and dried over MgSO<sub>4</sub>, concentrated in vacuo to afford oily residue which was purified over silica gel column chromatography (hexane:AcOEt = 1:1) to give 2.48 g of the compound (vi) as a colorless oil (Yield 80.6 %).

<sup>1</sup>H-NMR (CDCl<sub>3</sub>, δ): 7.97 (1H,s), 7.08-7.60 (25H,m), 4.91-5.19 (5H,m), 4.30 (2H,m), 0.89 (9H,s), 0.08 (6H,s)  
IR (neat, cm<sup>-1</sup>): 3060, 3040, 2950, 2925, 2880, 2855, 1517, 1492, 1475, 1380, 1350, 1310, 1285, 1215, 1070, 1060, 1130, 1085, 1010, 910, 878, 840, 780, 750, 700

(6) Dibenzyl-2-(hydroxy-2-(1-triphenylmethyl-(1,2,4)-triazol-3-yl)ethyl) phosphae: Compound (vii)

To a stirred solution of compound (vi) (777 mg, 1.04 mmol) dissolved in 7 ml of THF was added 1.1 ml of Bu<sub>4</sub>NF (1 M THF solution, 1.1 mmol). After the mixture was stirred at room temperature for 1 h, the reaction mixture was extracted with CHCl<sub>3</sub>. The combined extracts were washed with water brine, dried over MgSO<sub>4</sub> and concentrated to give an oily residue which was purified over silica gel column chromatography (CH<sub>2</sub>Cl<sub>2</sub>:MeOH =

10:1) to give 215 mg of the compound (vii) as a colorless oil (Yield 32.7 %).

$^1\text{H-NMR}$  ( $\text{CDCl}_3$ ,  $\delta$ ): 7.90 (1H,s), 7.02-7.60 (25H,m), 5.05 (2H,s), 5.00 (1H,m), 4.96 (2Hs), 4.33 (3H,m)

IR ( $\text{NaCl}$ ,  $\text{cm}^{-1}$ ): 3350, 3075, 3025, 1495, 1445, 1270, 1215, 1175, 1090, 1020, 910, 880, 750, 700

(7) 2-Hydroxy-2-(1H-(1,2,4)-triazol-3-yl)ethyl phosphate: Compound (viii)

To a stirred solution of compound (vii) (310 mg, 0.49 mmol) dissolved in 3 ml of  $\text{CH}_2\text{Cl}_2$  was added 0.26 ml (1.97 mmol) of  $\text{Me}_3\text{SiBr}$  at room temperature and the mixture was stirred for 1 h. Then 0.05 ml of EtOH was added and the mixture was stirred for another 1 h. To the solution were added propylene oxide (0.5 ml) and 5 ml of ether and the mixture was stirred at  $0^\circ\text{C}$  for 1 h. The gummy precipitates were collected, washed with ether and acetone, and dried in vacuo to give 50 mg of the compound (viii) (Yield 48.8 %).

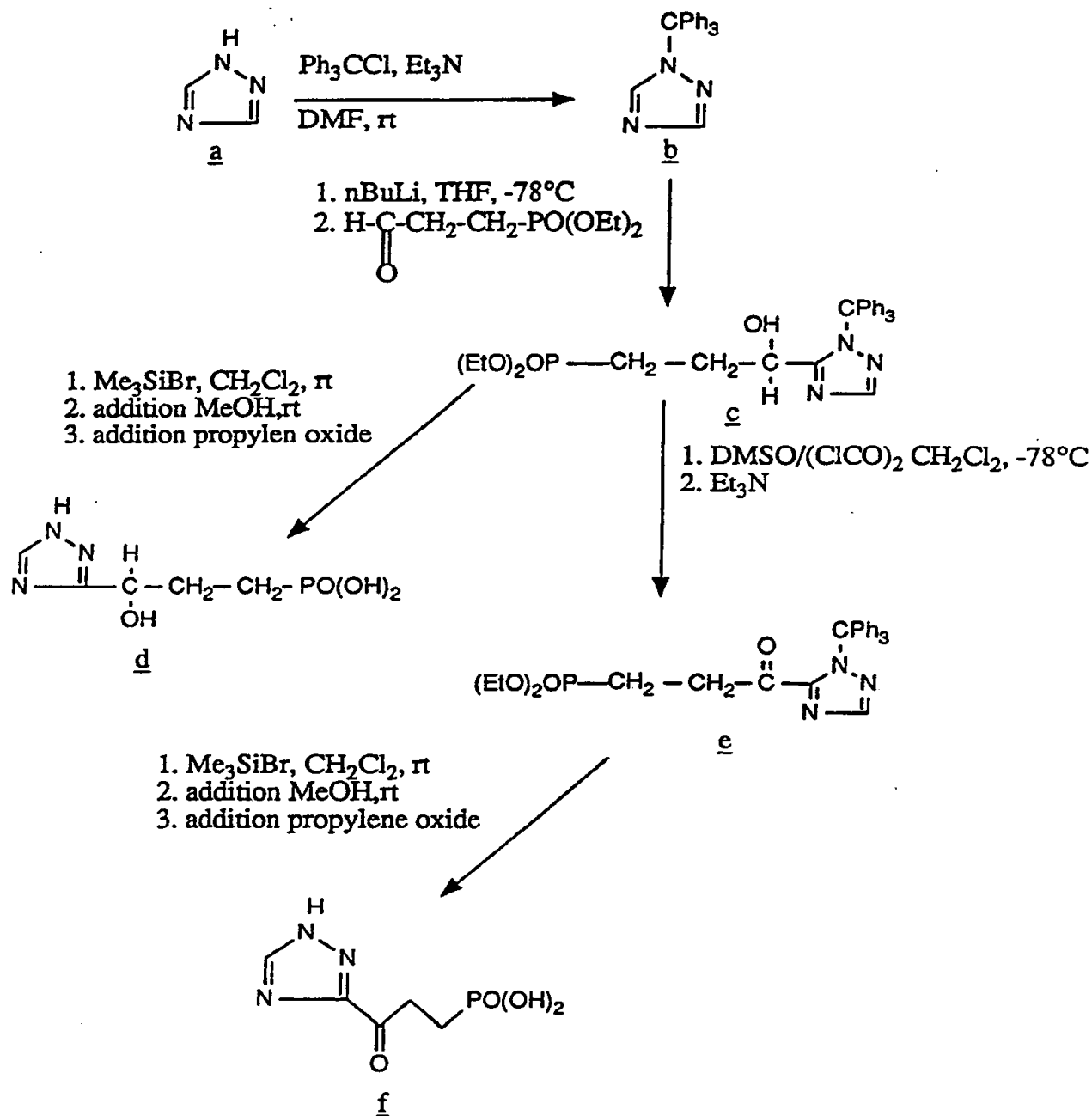
$^1\text{H-NMR}$  ( $\text{D}_2\text{O}$ ,  $\delta$ ): 9.17 (1H,s), 5.25 (1H,t,J=4.0 Hz), 4.15 (2H,dd,J=7.0 Hz)

The thus obtained compound corresponds to the substance of compound No. 106 in table 1.

Example 3

A phosphonate compound of this invention was prepared according to the flow charts shown below:

- 33 -



(1) Diethyl 3-(1-triphenylmethyl-(1,2,4)-triazol-5-yl)-3-hydroxypropyl-phosphonate:  
Compound c

(1,2,4)-Triazole (Compound a) was protected with chlorotriphenylmethane to afford 1-triphenylmethyl-(1,2,4)-triazol (Compound b), and 4.67 g (15.0 mmoles) of the thus obtained Compound b was dissolved in 150 ml of THF. The resulting solution was cooled to  $-78^\circ\text{C}$  and n-BuLi (1.5 M hexane solution, 16.5 mmoles) was added thereto dropwise

- 34 -

over 5 minutes. After the mixture was stirred at  $-78^{\circ}\text{C}$  for one hour, 2.67 ml (15.0 mmol) of an aldehyde prepared according to the process disclosed in DE 251634 was added thereto over 5 minutes, followed by stirring of the resulting mixture for 3 hours. The mixture was quenched to  $-78^{\circ}\text{C}$ , and 5 ml of an aqueous  $\text{NH}_4\text{Cl}$  was added thereto. After the temperature of the mixture was elevated to room temperature, the mixture was diluted with 100 ml of ethyl acetate, washed successively with 200 ml of water and 50 ml of brine, dried over  $\text{Na}_2\text{SO}_4$  and concentrated to give 5.72 g of a colorless solid. The solid was subjected to silica gel (175 g) chromatography using ethyl acetate and then  $\text{CH}_2\text{Cl}_2/\text{CH}_3\text{OH} = 19:1$  to obtain 3.17 g of the title compound as a colorless solid (Yield 42 %).

m.p.  $147$  to  $152^{\circ}\text{C}$

IR (NaCl)  $\text{cm}^{-1}$ : 3340, 1595, 1490, 1445, 1240, 1200, 1060, 1030, 960, 755, 700

$^1\text{H}$ -NMR (90 MHz,  $\text{CDCl}_3$ )  $\delta$ : 7.98 (1H, s), 7.05-7.53 (15H, m), 4.75 (1H, d,  $J=8.2$  Hz), 4.02 (4H, quintet,  $J=7.4$  Hz), 3.95-4.31 (1H, m), 1.50-2.10 (2H, m), 1.29 (6H, t,  $J=7.0$  Hz), 0.55-1.15 (2H, m)

Elementary analysis for  $\text{C}_{28}\text{H}_{32}\text{O}_4\text{N}_3\text{P}$

Calculated: C 66.52; H 6.38; N 8.31

Found C 66.57; H 6.46; N 8.02

(2) 3-(1H-1,2,4-triazol-3-yl)-3-hydroxypropyl-phosphonic acid: Compound d

To a solution of Compound c dissolved in 100 ml of  $\text{CH}_2\text{Cl}_2$  was added 5.28 ml (40.0 mmol) of  $\text{Me}_3\text{SiBr}$ , and the resulting mixture was stirred at room temperature overnight. To the mixture was added 25 ml of  $\text{CH}_3\text{OH}$ , and the resulting mixture was stirred at room temperature for one hour. Upon addition of 3.0 ml of propylene oxide, a gummy substance precipitated, which was dissolved by addition of  $\text{CH}_3\text{OH}$  (50 ml). When 400 ml of ether was slowly added to the solution, a micropowder precipitated. After stirring of the mixture at room temperature for one hour, the powder was filtered out, washed several times with ether and dried in vacuo to give 1.97 g of the title compound as a hygroscopic powder (Yield 95 %).

m.p.  $>60^{\circ}\text{C}$  (decomp.)

$^1\text{H}$ -NMR (90 MHz,  $\text{D}_2\text{O}$ )  $\delta$ : 8.85 (1H, s), 5.11 (1H, t,  $J=6.3$  Hz), 1.45-2.40 (4H, m)

The thus obtained compound corresponds to the substance of the Compound No. 1 in

Table 1.

(3) Compound e

To a solution of oxalyl chloride (0.69 ml, 7.91 mmoles) dissolved in 40 ml of methylene chloride was added dropwise 1.12 ml (15.8 mmoles) of dimethyl sulfoxide at  $-78^{\circ}\text{C}$  over 10 minutes. After the resulting mixture was stirred for 15 minutes, a solution of Compound c (2.0 g, 3.69 mmoles) dissolved in 10 ml of methylene chloride was added thereto over 10 minutes, followed by stirring at  $-78^{\circ}\text{C}$  for one hour. To the mixture was added 3.31 ml (23.8 mmoles) of triethylamine over 10 minutes, and the temperature of the mixture was elevated to room temperature over one hour. The reaction mixture was washed successively with 100 ml of water and brine, dried over  $\text{Na}_2\text{SO}_4$  and concentrated to give 2.01 g of a colorless solid, which was subjected to silica gel chromatography using ethyl acetate to give 1.10 g of the title compound as a colorless solid (Yield 55 %).

m.p. 167 to  $169^{\circ}\text{C}$

IR (NaCl)  $\text{cm}^{-1}$ : 2680, 1720, 1485, 1450, 1260, 1060, 1030, 960, 860, 801, 750, 700

$^1\text{H-NMR}$  (90 MHz,  $\text{CDCl}_3$ )  $\delta$ : 7.98 (1H, s), 6.91-7.40 (15H, m), 4.04 (4H, quintet,  $J=7.0$  Hz), 2.87-3.14 (2H, m), 1.45-1.76 (2H, m), 1.28 (6H, t,  $J=7.0$  Hz)

Elementary analysis for  $\text{C}_{28}\text{H}_{30}\text{O}_4\text{N}_3\text{P}$

Calculated: C 66.78; H 6.01; N 8.35

Found C 66.64; H 6.07; N 8.21

(4) 3-(1H-1,2,4-triazol-3-yl)-3-oxopropyl-phosphonic acid: Compound f

To a solution of Compound e (0.79 g, 1.57 mmoles) dissolved in 10 ml of methylene chloride was added 0.62 ml (4.71 mmoles) of trimethylsilyl bromide, and the resulting mixture was stirred at room temperature overnight. To the resulting mixture was added 6.0 ml of methanol, and the mixture was stirred for one hour. After addition of 2.0 ml of propylene oxid and stirring at room temperature for 15 minutes, 50 ml of ether was added to the mixture to effect crystallization of the desired product. The precipitate was collected, washed several times with ether and dried to give 0.34 g of the title compound as a white powder (Yield 100 %).

m.p. 196 to  $198^{\circ}\text{C}$ .

IR (NaCl)  $\text{cm}^{-1}$ : 1710, 1450, 1410

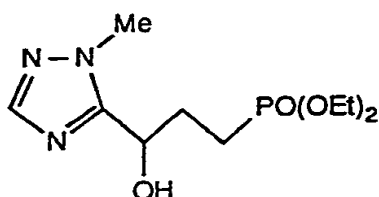
$^1\text{H-NMR}$  (90 MHz,  $\text{D}_2\text{O}$ )  $\delta$ : 8.66 (1H, s), 3.28-3.58 (2H, m), 1.98-2.35 (2H, m)

- 36 -

The thus obtained compound corresponds to the substance of the Compound No. 2 in Table 1.

Example 4:

Preparation of Diethyl 3-(1-Methyl-(1,2,4)-triazole-5-yl)-3-hydroxypropylphosphonate



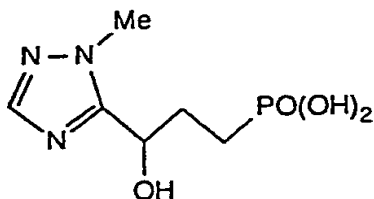
(Compound 101)

To a solution of 1-methyl-1,2,4-triazole (3.0 g, 36.1 mmol) in 60 ml of THF at  $-78^{\circ}\text{C}$  under nitrogen was added *n*-BuLi (43.3 mmol, 1.5 M in hexane). After stirring for 1 h, diethyl-3-oxopropylphosphate (8.4 g, 43.3 mmol) was added and stirred for an additional 1 h. The reaction was quenched with aqueous saturated  $\text{NH}_4\text{Cl}$  (ca. 15 ml) and extracted with ethyl acetate (3 x 150 ml). The aqueous layer was saturated with NaCl and extracted further with dichloromethane (3 x 100 ml). The combined organic layers were dried over  $\text{MgSO}_4$ , concentrated, and purified by silica gel chromatography ( $\text{AcOEt/EtOH}$ ) to give 3.60 g (30 % yield) of compound 101 as an oil.

$^1\text{H}$  NMR (90 MHz,  $\text{CDCl}_3$ )  $\delta$ : 7.78 (s, 1H), 5.00 (t, 1H,  $J=6.2$  Hz), 4.20 (quint, 4,  $J=7.3$ ), 4.00 (s, 3H), 2.50-1.60 (m, 5H), 1.34 (t, 6H,  $J=7.0$ ).

Example 5:

Preparation of (1-Methyl-(1,2,4)-triazole-5-yl)-3-hydroxypropylphosphonic acid



(Compound 102)

To a solution of compound 101 (2.0 g, 7.2 mmol) in 40 ml of dichloromethane was added

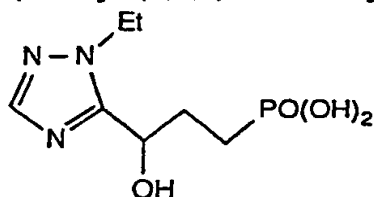
- 37 -

bromotrimethylsilane (4.74 mmol, 35.9 mmol) at room temperature under nitrogen and stirred for 17 h. After addition of 20 ml of methanol and stirring for 2 h, propylene oxide (2.0 ml) was added, stirred for 1 h, and diluted with ether. Precipitates were collected on a glass filter, washed with ether, and dried to give 1.0 g (63 % yield) of compound 102 as white powder. The melting point was obtained after lyophilizing from water.

mp; 64-74°C:  $^1\text{H-NMR}$  (90 MHz,  $\text{D}_2\text{O}$ )  $\delta$ : 8.39 (s, 1H), 5.23 (t, 1H,  $J=6.3$ ), 4.02 (s, 3H), 2.38-1.58 (m, 4H).

### Example 6:

Preparation of (1-Ethyl-(1,2,4)-triazol-5-yl)-3-hydroxypropylphosphonic acid



(Compound 103)

Compound 103 was prepared by the same procedure by starting with 1-ethyl-1,2,4-triazole instead of 1-methyl-1,2,4-triazole.

mp; 127-130°C:  $^1\text{H-NMR}$  (90 MHz,  $\text{D}_2\text{O}$ )  $\delta$ : 8.41 (s, 1H), 5.25 (t, 1H,  $J=6.5$ ), 4.40 (q, 2H,  $J=7.4$ ), 2.42-1.50 (m, 4H), 1.50 (t, 3H,  $J=7.4$ ).

1-Methyl and 1-ethyl-1,2,4-triazole were prepared according to the procedure by Dallacker and Minn.

Dallacker, F.; Minn, K. Chemiker-Zeitung, 1986, 110, 101-108

### Formulation examples of active substances of the formula (2)

(% = percent by weight)

#### 1. Emulsion concentrates

Active substance from Table 1

Ca dodecylbenzenesulfonate

Octylphenol polyethylene glycol ether

(4-5 mol of EO)

a) b)

10 %

3 %

3 %

1 %

3 %

3 %

- 38 -

Castor oil polyethylene glycol ether (36 mol of EO)	4 %	4 %
Cyclohexanone	30 %	10 %
Xylene mixture	50 %	79 %

Emulsions of any desired concentration can be prepared from such concentrates by diluting them with water.

<u>2. Suspension concentrate</u>	a)	b)
Active substance from Table 1	5 %	40 %
Ethylene glycol	10 %	10 %
Nonylphenol polyethylene glycol ether (15 mol of EO)	1 %	6 %
Na ligninsulfonate	5 %	10 %
Carboxymethylcellulose	1 %	1 %
37 % aqueous formaldehyde solution	0.2 %	0.2 %
Silicone oil in the form of a 75% aqueous emulsion	0.8 %	0.8 %
Water	77 %	32%

The finely-ground active substance is mixed intimately with the additives. This gives a suspension concentrate, from which suspensions of any desired concentration can be prepared by diluting it with water.

### 3. Salt solution

Active substance from Table 1	5 %
Isopropylamine	1 %
Octylphenol polyethylene glycol ether (78 mol of EO)	3 %
Water	91%

The compounds of the formula (2) are employed as such or preferably as compositions together with the auxiliaries customary in formulation technology, and they are therefore processed in a known manner to give, for example, emulsion concentrates, directly sprayable or dilutable solutions, dilute emulsions, sprayable powders, soluble powders, dusts, granules, and also encapsulations, for example in polymeric substances. The



application methods, such as spraying, atomising, dusting, scattering or pouring, as well as the type of compositions are selected to suit the intended aims and the prevailing circumstances.

### Biological Examples

#### Example B1: Herbicidal action before emergence of the plants

The test plants are seeded out in plastic pots containing standard soil. Immediately after seeding, the pots are being sprayed with an aqueous suspension of the compound No. 102. The rate corresponds to 4000g a.i./ha. The treated pots are then placed in the greenhouse at temperatures of 18°C (night) and 24°C (day). Appr. 3 weeks after treatment, the emerged plants are evaluated in terms of herbicidal symptoms:

- 1: plants have not emerged or are totally withered
- 2-3: very pronounced action
- 4-6: medium action
- 7-8: weak action
- 9: no action (as untreated controls).

In this test, the compound 102 given in Table 1 shows very pronounced herbicidal action against the weeds.

#### Example B2: Post-emergence herbicidal action (contact herbicide)

The test plants are seeded out in plastic pots containing standard soil and raised in the greenhouse at 18°C (night) and 24°C (day). Appr. 10 to 20 days after seeding (depending of individual growth-rate), foliar treatment takes place with an aqueous suspension of the compound No. 102. The rate corresponds to 2000g a.i./ha. Appr. 2 weeks after treatment, the emerged plants are evaluated in terms of herbicidal symptoms:

- 1: plants have not emerged or are totally withered
- 2-3: very pronounced action
- 4-6: medium action
- 7-8: weak action
- 9: no action (as untreated controls).

- 40 -

In this test, the compound 102 given in Table 1 shows very pronounced herbicidal action (rating "3") against the weed "Setaria" and medium action (rating "4") against the weed "Stellaria".

## CLAIMS

1. A compound represented by formula (1)



wherein  $\text{R}_1$  represents a hydrogen atom or a group A which is a protective group or is  $\text{C}_1$ - $\text{C}_4$ -alkyl; X and Y jointly represent a carbonyl group together with the carbon atom to which they are bonded, or X and Y each independently represent a hydrogen atom or a  $-\text{OR}_2$  group, wherein  $\text{R}_2$  represents a hydrogen atom, a lower alkyl group, an acetyl group, a benzyl group or a silyl group represented by  $-\text{SiR}'_3$  (wherein  $\text{R}'$  represents an alkyl group); and Z represents a  $-\text{CH}_2\text{PO}(\text{OR}_3)_2$  or  $-\text{CH}_2\text{OPO}(\text{OR}_3)_2$  group (wherein  $\text{R}_3$  represents a hydrogen atom, a lower alkyl group, a benzyl group, a phenyl group, a cyanoethyl group, an allyl group, a  $\text{CH}_2\text{OH}$  group, a  $\text{COOR}_5$  group, or an alkali metal, alkaline earth metal, ammonium, organic ammonium, trialkylsulphonium, trialkylsulfoxonium, phosphonium or amidinium cation;

n is 0 or 1; and

$\text{R}_5$  is a  $\text{C}_1$ - $\text{C}_6$ -alkyl group.

2. A compound of the formula (1) according to Claim 1, wherein

$\text{R}_1$  represents a hydrogen atom or a group A which is a protective group; X and Y jointly represent a carbonyl group together with the carbon atom to which they are bonded, or X and Y each independently represent a hydrogen atom or a  $-\text{OR}_2$  group, wherein  $\text{R}_2$  represents a hydrogen atom, a lower alkyl group, an acetyl group, a benzyl group or a silyl group represented by  $-\text{SiR}'_3$  (wherein  $\text{R}'$  represents an alkyl group); and Z represents a  $-\text{CH}_2\text{PO}(\text{OR}_3)_2$  or  $-\text{CH}_2\text{OPO}(\text{OR}_3)_2$  group (wherein  $\text{R}_3$  represents a hydrogen atom, a lower alkyl group, a benzyl group, a phenyl group, a cyanoethyl group or an allyl group or a  $\text{CH}_2\text{OH}$  group or a  $\text{COOR}_5$  group,

n is 1; and

$\text{R}_5$  is a alkyl group.

3. A compound according to claim 1 of the formula (2)

- 42 -



wherein  $\text{R}'_1$  represents a hydrogen atom or  $\text{C}_1$ - $\text{C}_4$ -alkyl; X and Y jointly represent a carbonyl group together with the carbon atom to which they are bonded, or X and Y each independently represent a hydrogen atom or a  $-\text{OR}_2$  group, wherein  $\text{R}_2$  represents a hydrogen atom, a lower alkyl group, an acetyl group, a benzyl group or a silyl group represented by  $-\text{SiR}'_3$  (wherein  $\text{R}'$  represents an alkyl group); and  $\text{Z}'$  represents a  $-\text{CH}_2\text{PO}(\text{OR}_3)_2$  or  $-\text{CH}_2\text{OPO}(\text{OR}_3)_2$  group (wherein  $\text{R}_3$  represents a hydrogen atom, a lower alkyl group, a benzyl group, a phenyl group, a cyanoethyl group, an allyl group, or an alkali metal, alkaline earth metal, ammonium, organic ammonium, trialkylsulphonium, trialkylsulfoxonium, phosphonium or amidinium cation; and  $n$  is 0 or 1.

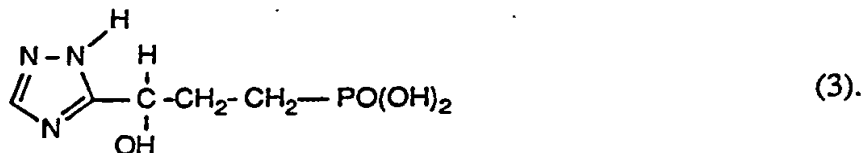
4. A compound according to claim 3, wherein

$\text{R}'_1$  represents a hydrogen atom; X and Y jointly represent a carbonyl group together with the carbon atom to which they are bonded, or X and Y each independently represent a hydrogen atom or a  $-\text{OR}_2$  group, wherein  $\text{R}_2$  represents a hydrogen atom, a lower alkyl group, an acetyl group, a benzyl group or a silyl group represented by  $-\text{SiR}'_3$  (wherein  $\text{R}'$  represents an alkyl group); and  $\text{Z}$  represents a  $-\text{CH}_2\text{PO}(\text{OR}_3)_2$  or  $-\text{CH}_2\text{OPO}(\text{OR}_3)_2$  group (wherein  $\text{R}_3$  represents a hydrogen atom, a lower alkyl group, a benzyl group, a phenyl group, a cyanoethyl group, an allyl group, or an alkali metal, alkaline earth metal, ammonium, organic ammonium, trialkylsulphonium, trialkylsulfoxonium, phosphonium or amidinium cation; and  $n$  is 0 or 1.

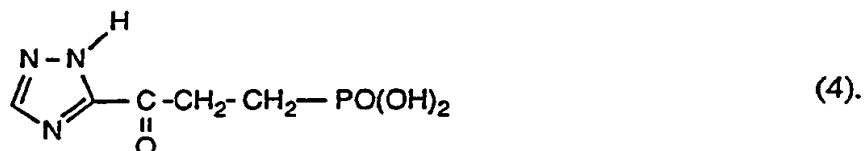
5. A compound according to Claim 3, wherein X and Y independently represent a  $-\text{OR}_2$  group, wherein  $\text{R}_2$  represents an acetyl group.

6. A compound according to Claim 3, wherein the compound is represented by formula (3):

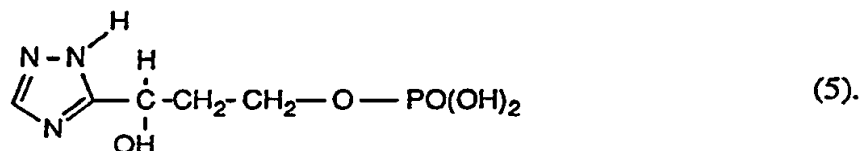
- 43 -



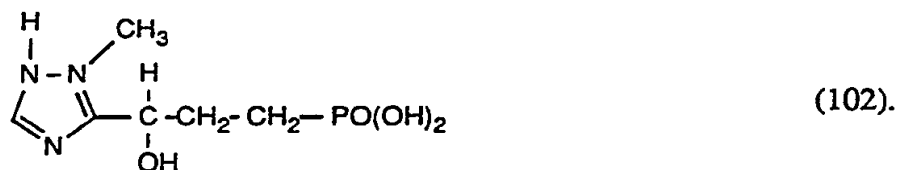
7. A compound according to Claim 3, wherein the compound is represented by formula (4):



8. A compound according to Claim 3, wherein the compound is represented by formula (5):



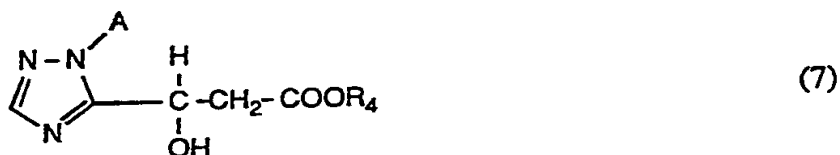
9. A compound according to Claim 3, wherein the compound is represented by formula (102):



10. A compound according to Claim 1, wherein the protective group A represents a triphenylmethyl group, a benzyl group, a tert-butoxycarbonyl group, an allyl group or a sulfonyl group.

11. A process for producing the compound represented by formula (1) as claimed in Claim 1, wherein n is 1 and Z represents  $-\text{CH}_2\text{OPO}(\text{OR}_3)_2$ , which comprises: forming a ((1,2,4)-triazol-5-yl)propionic acid ester represented by formula (7):

- 44 -



wherein  $R_4$  represents an alkyl group; and A has the same meaning as defined below,

from a ((1,2,4)-triazol-5-yl)aldehyde represented by formula (6):

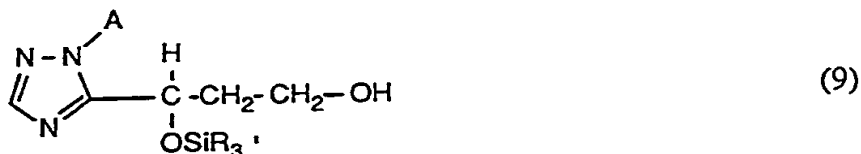


wherein A represents a protective group;

reacting the ((1,2,4)-triazol-5-yl)propionic acid ester thus formed with a suitable alkylsilyl halide to form a 3-alkylsilyloxy-3-((1,2,4)-triazol-5-yl)propionic acid ester represented by formula (8):



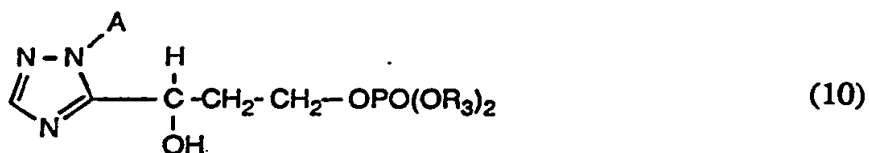
wherein  $R'$  represents an alkyl group; and A and  $R_4$  have the same meanings as defined above, reducing the 3-alkylsilyloxy-3-((1,2,4)-triazol-5-yl)-propionic acid ester to form a 3-alkylsilyloxy-3-((1,2,4)-triazol-5-yl)-1-propanol represented by the formula (9):



wherein  $R'$  and A have the same meanings as defined above;

reacting the 3-alkylsilyloxy-3-((1,2,4)-triazol-5-yl)-1-propanol with a suitable phosphine

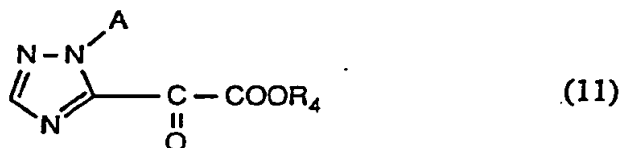
compound to form a 3-hydroxy-3-((1,2,4)-triazol-5-yl)propyl-phosphate represented by formula (10):



wherein  $R_3$  has the same meaning as defined in Claim 1; and A has the same meaning as defined above; and converting, as necessary, the 3-hydroxy-3-((1,2,4)-triazol-5-yl)propyl-phosphate to 3-hydroxy-3-((1,2,4)-triazol-5-yl)propyl-phosphate (mono- or tri-ester form).

12. A process for producing the compound represented by the formula (1) as claimed in claim 1, wherein n is 0 and Z represents  $-\text{CH}_2\text{OPO}(\text{OR}_3)_2$ , which comprises:

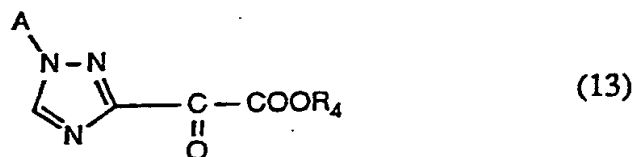
forming a 2-((1,2,4)-triazol-5-yl)-2-oxoacetic acid ester represented by formula (11):



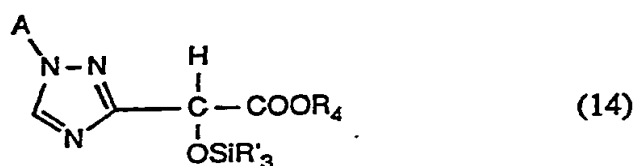
wherein  $R_4$  represents an alkyl group; and A represents a group which is a protective group or is  $\text{C}_1\text{--C}_4$ -alkyl, from a (1,2,4)-triazole represented by formula (12):



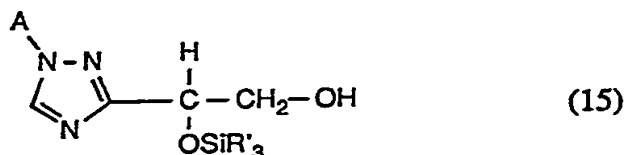
wherein A represents a group which is a protective group or is  $\text{C}_1\text{--C}_4$ -alkyl; isomerizing the 2-((1,2,4)-triazol-5-yl)-2-oxoacetic acid ester thus formed to form 2-((1,2,4)-triazol-3-yl)-2-oxoacetic acid ester represented by formula (13):



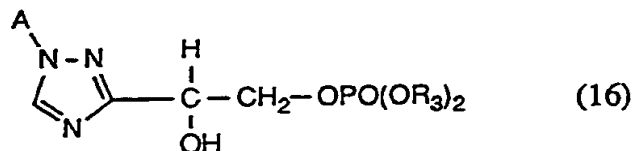
wherein A and R<sub>4</sub> have the same meanings as defined above; reducing the 2-((1,2,4)-triazol-3-yl)-2-oxoacetic acid ester followed by silylation to form 2-((1,2,4)-triazol-3-yl)-2-alkylsilyloxy-acetic acid ester represented by formula (14):



wherein R' represents an alkyl group; and A and R<sub>4</sub> have the same meanings as defined above, reducing the 2-alkylsilyloxy-2-((1,2,4)-triazol-3-yl)acetic acid ester to form a 2-alkylsilyloxy-2-((1,2,4)-triazol-3-yl)-1-ethanol represented by formula (15):



wherein R' and A have the same meanings as defined above; reacting the 2-alkylsilyloxy-2-((1,2,4)-triazol-3-yl)-1-ethanol with a suitable phosphine compound followed by desilylation to form a 2-hydroxy-2-((1,2,4)-triazol-3-yl)ethyl-phosphate represented by formula (16):



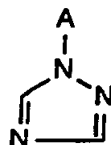
wherein A and R<sub>3</sub> have the same meaning as defined above; and converting, as necessary, the 2-hydroxy-2-((1,2,4)-triazol-3-yl)ethyl-phosphate to 2-hydroxy-2-((1,2,4)-triazol-3-yl)-ethyl-phosphate (mono or triester form).

13. A process for producing the compound represented by the formula (1) as claimed in



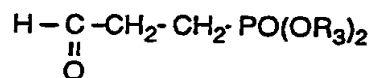
Claim 1, wherein n is 1 and Z represents  $-\text{CH}_2\text{PO}(\text{OR}_3)_2$ , which comprises:

reacting a (1,2,4)-triazole represented by formula (12):



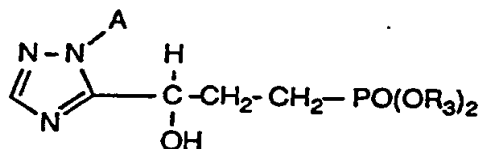
(12)

wherein A represents a group which is a protective group or is  $\text{C}_1$ - $\text{C}_4$ -alkyl; with an aldehyde compound represented by the formula (23):



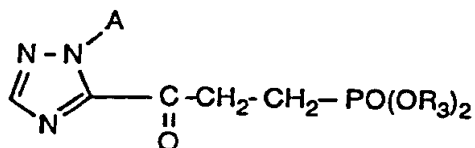
(23)

wherein  $\text{R}_3$  has the same meaning as defined above, to form a 3-((1,2,4)-triazol-5-yl)-3-hydroxypropyl-phosphonate represented by formula (24):



(24)

wherein  $\text{R}_3$  and A have the same meanings as defined above; converting the 3-((1,2,4)-triazol-5-yl)-3-hydroxypropyl-phosphonate to 3-(1H-1,2,4-triazol-5-yl)-3-hydroxypropyl-phosphonic acid or phosphonate or oxidizing said 3-((1,2,4)-triazol-5-yl)-3-hydroxypropyl-phosphonate to form a 3-((1,2,4)-triazol-5-yl)-3-oxopropyl-phosphonate represented by formula (25)



(25)

wherein  $\text{R}_3$  and A have the same meanings as defined in Claim 1, followed by conversion into 3-(1H-1,2,4-triazol-5-yl)-3-oxopropyl-phosphonic acid or phosphonate.

14. A herbicidal composition comprising as the active ingredient at least one of the compounds represented by formula (2) as claimed in Claim 3.

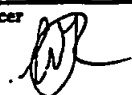
15. A fungicidal composition comprising as the active ingredient at least one of the compounds represented by formula (2) as claimed in Claim 3.

16. A method of controlling undesired plant growth, which comprises applying an effective amount of an active substance of the formula (2) according to claim 3, or a composition comprising this active substance according to claim 14, to the plants or their environment.

## INTERNATIONAL SEARCH REPORT

PCT/JP 92/00485

International Application

<b>I. CLASSIFICATION OF SUBJECT MATTER</b> (If several classification symbols are indicated, indicate all) <sup>6</sup>		
According to International Patent Classification (IPC) or to both National Classification and IPC Int.Cl. 5 C07F9/6518; A01N57/00		
<b>II. FIELDS SEARCHED</b>		
Minimum Documentation Searched <sup>7</sup>		
Classification System	Classification Symbols	
Int.Cl. 5	C07F ; A01N	
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched <sup>8</sup>		
<b>III. DOCUMENTS CONSIDERED TO BE RELEVANT<sup>9</sup></b>		
Category <sup>10</sup>	Citation of Document, <sup>11</sup> with indication, where appropriate, of the relevant passages <sup>12</sup>	Relevant to Claim No. <sup>13</sup>
A	EP,A,0 078 613 (IMPERIAL CHEMICAL INDUSTRIES PLC) 11 May 1983 cited in the application see the whole document ---	1-16
A	EP,A,0 065 216 (HOECHST AG) 24 November 1982 see the whole document ---	1-16
A	EP,A,0 275 821 (CIBA-GEIGY AG) 27 July 1988 see the whole document ---	1-13
<p><sup>10</sup> Special categories of cited documents :</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"&amp;" document member of the same patent family</p>		
<b>IV. CERTIFICATION</b>		
Date of the Actual Completion of the International Search 15 JUNE 1992		Date of Mailing of this International Search Report 09. 07. 92
International Searching Authority EUROPEAN PATENT OFFICE		Signature of Authorized Officer RINKEL L. J. 

Form PCT/ISA/210 (second sheet) (January 1985)

# ANNEX TO THE INTERNATIONAL SEARCH REPORT ON INTERNATIONAL PATENT APPLICATION NO.

JP 9200485  
SA 58435

This annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report.  
The members are as contained in the European Patent Office EDP file on  
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information. 15/06/92

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP-A-0078613	11-05-83	AU-B- 550022	27-02-86
		AU-A- 8961182	12-05-83
		CA-A- 1179679	18-12-84
		GB-A- 2114133	17-08-83
		JP-A- 58088397	26-05-83
		US-A- 4439428	27-03-84
EP-A-0065216	24-11-82	DE-A- 3118257	02-12-82
		JP-A- 58023700	12-02-83
EP-A-0275821	27-07-88	AU-B- 607722	14-03-91
		AU-A- 8145387	26-05-88
		DE-A- 3776880	02-04-92
		JP-A- 63150291	22-06-88
		US-A- 4939130	03-07-90
		ZA-A- 8708698	23-05-88

EPO FORM P0079

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82

09/807236  
(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property Organization  
International Bureau



(43) International Publication Date  
4 May 2000 (04.05.2000)

PCT

(10) International Publication Number  
WO 00/24867 A3

(51) International Patent Classification<sup>7</sup>: C12N 9/90,  
15/61, 15/66, 15/82, 15/10, 15/11, C12Q 1/68, C12N  
15/70, 15/74, 15/80, 15/83, 5/04, A01H 1/00

(74) Agent: FEULNER, Gregory, J.; E.I. du Pont de Nemours  
and Company, Legal Patent Records Center, 1007 Market  
Street, Wilmington, DE 19898 (US).

(21) International Application Number: PCT/US99/24698

(22) International Filing Date: 21 October 1999 (21.10.1999)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:  
60/105,409 23 October 1998 (23.10.1998) US

(71) Applicant (for all designated States except US): E.I. DU  
PONT DE NEMOURS AND COMPANY [US/US]; 1007  
Market Street, Wilmington, DE 19898 (US).

(81) Designated States (national): AE, AL, AU, BA, BB, BG,  
BR, CA, CN, CR, CU, CZ, DM, EE, GD, GE, HR, HU, ID,  
IL, IN, IS, JP, KP, KR, LC, LK, LR, LT, LV, MG, MK, MN,  
MX, NO, NZ, PL, RO, SG, SI, SK, SL, TR, TT, UA, US,  
UZ, VN, YU, ZA.

(84) Designated States (regional): ARIPO patent (GH, GM,  
KE, LS, MW, SD, SL, SZ, TZ, UG, ZW), Eurasian patent  
(AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent  
(AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU,  
MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM,  
GA, GN, GW, ML, MR, NE, SN, TD, TG).

Published:

— with international search report

(88) Date of publication of the international search report:  
20 September 2001

For two-letter codes and other abbreviations, refer to the "Guid-  
ance Notes on Codes and Abbreviations" appearing at the begin-  
ning of each regular issue of the PCT Gazette.

(72) Inventors; and

(75) Inventors/Applicants (for US only): ABELL, Lynn,  
Marie [US/US]; 5 Laurel Court, Wilmington, DE 19808  
(US). ALLEN, Stephen, M. [US/US]; 2225 Rosewood  
Drive, Wilmington, DE 19810 (US). CAHOON, Rebecca,  
E. [US/US]; 2331 West 18th Street, Wilmington, DE  
19806 (US). TAO, Yong [CN/US]; 101-8 Thorn Lane,  
Newark, DE 19711 (US).

RECEIVED

JAN 17 2002

TECH CENTER 1600/2900

(54) Title: PLANT HISTIDINE BIOSYNTHETIC ENZYMES

(57) Abstract: This invention relates to an isolated nucleic acid fragment encoding a histidine biosynthetic enzyme. The invention also relates to the construction of a chimeric gene encoding all or a portion of the histidine biosynthetic enzyme, in sense or antisense orientation, wherein expression of the chimeric gene results in production of altered levels of the histidine biosynthetic enzyme in a transformed host cell.

WO 00/24867 A3

